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FORMULATION OF NEW MODIFIED ALKYD RESINS AND THEIR APPLICATION IN THE FIELD OF SURFACE COATINGS

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ABSTRACT

Alkyd resins are considered modified polyester resins, which may be modified with many chemicals such as polybasic acids, polyhydric alcohols and/or fatty acids to produce resins with different properties and give the final surface coating formulations wide performance range. In this research, the soya bean based N,N-bis(2-hydroxyethyl)thiophene-2-carboxamide (HETCA)-modified alkyd resins and N,N-bis(2-hydroxyethyl)furan-2-carboxamide (HEFCA)-modified alkyd resins are prepared by partial replacement of glycerol (G) with each (HETCA) (I) and (HEFCA) (III), respectively, which are reacted as source of polyhydric alcohols, with both phthalic anhydride (PA) and soya bean oil fatty acids (SOFA), as source of polyacids.

The new HETCA-modified alkyd resins and previously prepared HEFCA-modified alkyd resins enter in final coating formulations and their performance are evaluated and compared using the international standard test methods (ASTM) and involved the measurement of phyisco-mechanical properties such as viscosity, drying time, specular gloss, pencil hardness, adhesion, flexibility. In general, the results show that the modification enhances both phyisco-mechanical and chemical properties. The corrosion resistance of alkyd resins improved by such HETCA and HEFCA- modification

Additionally, expected corrosion inhibitors such as morpholino(thiophen-2-yl)methanone (II) and morpholino(furan-2-yl) methanone (IV) are prepared and their anticorrosion effect is investigated by two ways. First, as an additive with unmodified (commercial) alkyd resin in zinc phosphate primer formulations. Second, as an additive with previously prepared HETCA- and HEFCA-modified alkyd resins in the same primer formulations. Both of corrosion inhibitors (II) and (IV) improve the corrosion resistance at certain dosages.

Keywords— anticorrosion coating, corrosion inhibitors, modified alkyd resin, N,N-bis(2-hydroxyethyl)thiophene-2-carboxamide (HETCA), N,N-bis(2-hydroxyethyl)furan-2-carboxamide (HEFCA), morpholino(thiophen-2-yl)methanone, morpholino(furan-2-yl)methanone

INTRODUCTION

In the ongoing shift from solvent- to waterbased coatings, alkyds have stubbornly resisted the change, and for valid reasons. Solvent-based alkyds are cost efficient and versatile, with a long history of proven performance in architectural, industrial and specialty applications. They offer excellent adhesion, hardness, gloss and corrosion resistance. These resins are also highly viscous: they originate as solids, making it very difficult to formulate shelf-stable coatings without the addition of solvent. Despite numerous options, including water-reducible alkyds, modified alkyd dispersions and alkyd emulsions, only 10% of alkyd-based coatings are currently waterborne⁽¹⁾. Challenges to widespread adoption include delayed hardness development, lower gloss and reduced corrosion protection.

N,N-bis(2-hydroxyethyl)furan-2-carboxamide (HEFCA)-modified alkyd resins are previously prepared by partial replacement of glycerol with N,N-bis(2-hydroxyethyl)furan-2-carboxamide (HEFCA) which was reacted with Linseed oil fatty acids (LOFA) and phthalic anhydride (PA) without affecting the resin constant. The biological effect of HEFCA-modified alkyd resins are investigated and showed promising results⁽²⁾.

The modification of short and medium alkyd resins using perfluorinated urethane toluene isocyanate (PFUTI), and incorporates the modified resins in a set of paint formulations containing different ratios of zinc phosphate as an inhibitive pigment, achieves promising results and illustrates corrosion protective properties in various paint formulations and also in the paint formulation free of the inhibitive pigment, s qhows enhancement of the corrosion protection efficiencies by the improvement in the hydrophobicity of alkyd resin modified with PFUTI. In view of these results, minimizes or neglects the most expensive inhibitive pigments from an economical stand point⁽³⁾.

The usage of maleic anhydride as a partial replacement of phthalic anhydride in long oil modified alkyd resins is investigated. The usages of maleic anhydride as a partial replacement of phthalic anhydride at these resins as well as at the vegetable oils improve the characteristics of the film of paint, such as its hardness and its resistance against atmospheric and corrosive agents. Maleic anhydride influences also in the condensation time of alkyd resins. Usually, the maleic anhydride is added before the phthalic anhydride, because it links with the double bonds of the radicals of vegetable oils, raising the stereo chemical structure and helps to achieve, in a short time, the proper viscosity of the resin. Experiments have shown that the optimal quantity of maleic anhydride for achieving the desirable results is $2\%^{(4)}$.

Alkyd resins are recently prepared by using recycled poly (ethylene terephthalate), PET, and different vegetable oils. Glycolysis of PET waste using pentaerytheritol (PEr) was used to produce suitable hydroxyl oligomers, GPEr, for alkyd resin. The glycolysis was carried out in the presence of manganese acetate as a catalyst and m-cresol as a solvent at 220 °C. Alkyd resins were prepared using phthalic anhydride, PET waste, glycerin, sunflower oil or linseed oil and ethylene glycol to produce PET-based alkyd resin. GPEr was used instead of PET waste to produce the second type of alkyd resin based on GPEr. The reactions were carried out in the presence of butylhydroxytin oxide as a catalyst to prepare alkyd resins. The curing characteristics of the resins produced were investigated. Corrosion resistivity based on salt spray and cathodic disbondment were evaluated for the cured alkyd as organic coating for carbon steel⁽⁵⁾.

Recently microbial activity of N,N Di-hydroxyethyl-2-thiophenamide (HETCA) have been discussed and the studies showed promising results as biocidal coatings. Preparation of modified polyesteramides were also considered and found also improved the film performance and durability and lead to substantial antimicrobial growth control^(6,7).

Experimental Materials

The Soya bean fatty acid (SOFA) used is supplied by Guangrao Welfare Resin Factory, China, the phthalic anhydride (PA) and the diethanolamine (DEA) are obtained from Sdfine Indian. The Glycerin are obtained from El Gomhouria Co., Egypt. Thiophene -2-carboxylic acid is obtained from SIGMA Chemical Co., USA. Furan-2-carboxylic acid is obtained from SIG-MA-ALDRICH Co., United Kingdom.

METHODS AND TECHNIQUES

1. Preparation of N,N-bis(2-hydroxyethyl) thiophene-2-carboxamide (HETCA) I ^(6,7)

A mixture of thiophene -2-carboxylic acid (0.1 mole = 12.815 gm), freshly distilled diethanolamine (0.1 mole = 10.5 gm), and 21 ml xylene as a solvent were placed in 250 ml round-bottomed flask fitted with Dean and Stark apparatus, The reaction was allowed to reflux until the theoretical amount of water (0.1 mole = 1.8 gm) or 1.8 ml)was collected. The mixture was allowed to cool where a clear pale yellow viscous material was obtained.



2- Preparation of N,N-bis(2-hydroxyethyl) furan-2-carboxamide (HEFCA) III ⁽²⁾

A mixture of furan-2-carboxylic acid (0.1 mole = 11.208 gm), freshly distilled diethanolamine (0.1 mole=10.5 gm), and 21 ml xylene as a solvent were placed in 250 ml round-bottomed flask fitted with Dean and Stark apparatus, The reaction was allowed to reflux until the theoretical amount of water (0.1 mole = 1.8 gm or 1.8 ml) was collected. The mixture was allowed to cool where a brown viscous material was obtained.

Equation :

N,N-bis(2-hydroxyethyl)thiophene-2-carboxamide

HETCA



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3- Preparation Of Alkyd Resins via solvent process⁽⁸⁾:

The preparation of various modified alkyd resins carried out as shown in the following equations:

By partial replacement of Glycerol (G) with each of following compounds:

N,N-bis(2-hydroxyethyl)thiophene-2-carboxamide (HETCA) I

N,N-bis(2-hydroxyethyl)furan-2-carboxamide (HEFCA) III

As the ingredient source of the polyol, while Phthalic anhydride (PA) and Soya bean oil fatty acid (SOFA) is the source of the polybasic acid, in the presence of 10% xylene of their volume, the whole ingredient was placed in 250 ml roundbottomed flask fitted with Dean Stark apparatus.

It should be noted that within each set of formulation, the total number of acid and hydroxyl equivalents for various runs were kept constant and the partial replacement of Glycerol with each compound " I" and " III" is listed in table (1).

Table (1): List of hydroxyl equivalent of different rune

1 411		
Resin	G	HETCA or HEFCA
B _x	1.00	0.00
T_{x1} or F_{x1}	0.95	0.05
T_{χ_2} or F_{χ_2}	0.90	0.10
T_{x3} or F_{x3}	0.80	0.20
T _{x4} or F _{x4}	0.70	0.30

X= 0, 1, 2 or 3 at 0%, 10%, 20% or 30% excess-**OH respectively**

B: Unmodified Alkyd Resin (Blank)

T: HETCA-modified Alkyd Resin

F: HEFCA-modified Alkyd Resin

The following tables (2-3) illustrate the resin characteristic constants for various modified alkyd resin formulations.

These formulations cover a wide range of various excess hydroxyl contents (0%, 10%, 20%, and 30% excess-OH).It is important to mention that in each series of resins there are:

1- Blank formulation containing no modifier compound and is introduced for comparison purpose. This formulation will be B0, B1, B2 or B3 to indicate excess-OH of 0%, 10%, 20% or 30% respectively.

2- Gradual increase in the percentage of the modifier. The modifier is partially replaces the hydroxyl equivalent of Glycerol used in formulation. Such replacement is based on the hydroxyl equivalent constant and the second subscript numbers 1, 2, 3 and 4 are given after the formulation numbers to indicate 5%, 10%, 20% and 30% modification partial replacement, respectively.

3.1. Preparation of Unmodified Alkyd Resins⁽⁸⁾







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Table (2): Resin constants for HETCA	-modified alkyd resins

Resin No.	Excess –	Ingredients	E	F	e _o	e _A	e _B	R	H ₂ O off (ml)
	OH (%)								
		G	30.7	3	- 0 520		0.520		
		HETCA	107.6	2	0.520		0.520	1.00	67
01-04	0	FA	280	1	0.222	0.510		1.00	0.7
		PA	74.1	2	0.297	0.519			
		G	30.7	3	- 0 600	0.546	0.600		
- -	10	HETCA	107.6	2	- 0.600			C7	
11-14	10	FA	280	1	0.200			1.10	0.7
		PA	74.1	2	0.346	0.546			
		G	30.7	3	0.710		0.716		C 0
	20	HETCA	107.6	2	- 0.716			1.20	
21-24	20	FA	280	1	0.164	0.506		1.20	0.8
		PA	74.1	2	0.432	0.596			
		G	30.7	3	0.070	- 0.679	0.879	7.4	
-	20	HETCA	107.6	2	- 0.879				
31-34	30	FA	280	1	0.111			1.30	1.30 7.1
		PA	74.1	2	0.568				

G:glycerol; HETCA:N,N-bis(2-hydroxyethyl)thiophene-2-carboxamide; PA: Phthalic Anhydride; FA: Soya bean fatty acid E: Equivalent Weight; e_A : Number of acid equivalent ; e_B : Number of hydroxyl equivalent; e_0 : Total equivalent present at the start of the reaction; F: Functionality ; R:Ratio of total-OH groups to total-COOH groups (= e_B/e_A) ; H2O off (ml): Number of water millimeters at which the reaction will be Completed (= $e_{acid} \times 18 + e_{anhydride} \times 9$)

3.2. Preparation Of HEFCA-Modified Alkyd Resins Based On Soyabean Oil Fatty Acid⁽²⁾



4-Preparation of morpholino(thiophen-2yl)methanone II

A mixture of thiophene-2-carboxylic acid (0.1 mole = 12.815 gm), freshly distilled diethanolamine (0.1 mole=10.5 gm), and 21 ml xylene as a solvent were placed in 250 ml round-bottomed flask fitted with Dean and Stark apparatus, The reaction was allowed to reflux until the theoretical amount of water (0.2 mole = 3.6gm or 3.6 ml)was collected. The mixture was allowed to cool where a brown viscous material was obtained.

Equation :



Morpholino(thiophen-2-yl)methanone

Resin No.	Excess -OH (%)	Ingredients	Е	F	e ₀	e _A	e _B	R	H ₂ O off (ml)
		G	30.7	3	0.520		0.520		
т	0	HEFCA	99.5	2	0.520		0.520	1.00	(7
I ₀₁₋₀₄	0	FA	280	1	0.222	0.510		1.00	0./
		PA	74.1	2	0.297	0.519			
		G	30.7	3	0.600		0.000		
T	10	HEFCA	99.5	2	0.600		0.600	0.600	< 7
I 11-14	10	FA	280	1	0.200	0.546		1.10	6./
		PA	74.1	2	0.346	0.546			
		G	30.7	3	0.71(0.716		
T	20	HEFCA	99.5	2	0./16		0./16	1.00	()
I ₂₁₋₂₄	20	FA	280	1	0.164	0.506		1.20	6.8
		PA	74.1	2	0.432	0.596			
		G	30.7	3	0.070		0.070		
T	20	HEFCA	99.5	2	0.8/9		0.879	1.20	7.1
1 ₃₁₋₃₄	30	FA	280	1	0.111	0.670		1.30	/.1
		РА	74.1	2	0 568	0.679			

FORMULATION OF NEW MODIFIED ALKYD RESINS AND THEIR APPLICATION 27 Table (3): Resin constants for HEFCA -modified alkyd resins

G:glycerol; HEFCA:N,N-bis(2-hydroxyethyl)furan-2-carboxamide; PA: Phthalic Anhydride; FA: Soya bean fatty acid E: Equivalent Weight; e_A : Number of acid equivalent ; e_B : Number of hydroxyl equivalent; e_0 : Total equivalent present at the start of the reaction; F: Functionality ; R:Ratio of total-OH groups to total-COOH groups (= e_B/e_A) ; H2O off (ml): Number of water millimeters at which the reaction will be Completed (= $e_{acid} \times 18 + e_{anhydride} \times 9$)

5-Preparation of furan-2-yl(morpholino) methanone IV

A mixture of furan-2-carboxylic acid (0.1 mole = 11.208 gm), freshly distilled diethanolamine (0.1 mole=10.5 gm), and 21 ml xylene as a solvent were placed in 250 ml round-bottomed flask fitted with Dean and Stark apparatus, The reaction was allowed to reflux until the theoretical amount of water (0.2 mole = 3.6 gm or 3.6 ml)was collected. The mixture was allowed to cool where a brown viscous material was obtained.

Equation :

Furan-2-yl(morpholino)methanone

Spectral analysis of prepared compounds:



Structural features of each of the previous compounds is confirmed by IR spectroscopy which tabulated in table (4, 5, 6 and 7) and Figures (1, 2, 3 and 4) respectively **Table (4): FTIR spectra of N,N-bis (2-hydroxy-**

ethyl) thiophene-2-carboxamide (HETCAI

Functional group	IR peak (cm ⁻¹)
ОН	3363 (Broad band)
CH aromatic	3099
CH aliphatic	2955
CO amide	1705





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Table (5): Spectral analysis of morpholino (thiophen-2-yl) methanone II

Functional group	IR peak (cm ⁻¹)
ОН	diminished
CH aromatic	3097
CH aliphatic	2943
CO amide	1708



Thu Nov 03 12:37:22 2016 (GMT-02:00) Figure (2) FTIR spectra of morpholino(thiophen-2-yl)methanone II

1001 917.5

Table (6): Spectral analysis of N,Nbis(2-hydroxyethyl)furan-2-carboxamide (HEFCA) III

Functional group	IR peak (cm ⁻¹)
	3391(Broad
OH	band)
CH aromatic	3050
CH aliphatic	2957
CO amide	1719

95 90 85 80 75 60 55 60 55 60 55 40 35 20 15 10 5 2000 mbers (cm-1)

Figure (3) I.R. Spectrum of N,N-bis(2-hydroxyethyl)furan-2-carboxamide (HEFCA) III

Table (7): Spectral analysis of furan-2-yl(morpholino)methanone IV

Functional group	IR peak (cm ⁻¹)
ОН	diminished
CH aromatic	3024
CH aliphatic	2955
CO amide	1734

85 80 70 50 v 03 10:52:26 2016 (GMT+02:00)

Method of testing and evaluation:

The structure of the prepared N,N-bis(2-hydroxyethyl)thiophene-2-carboxamide (HETCA) I, morpholino(thiophen-2-yl)methanone II. N,N-bis(2-hydroxyethyl)furan-2-carboxamide (HEFCA) III and furan-2-yl(morpholino)methanone IV was confirmed by FTIR and showed in Figure 1, 2 3 and 4 respectively.

All resins are adjusted to solid content = 60% (according to ASTM Method, D1644 - 01 (Reapproved 2012) and the following tests were carried out according to international standards:

Test method for color of transparent liquid, ASTM method, D1544 - 04 (Reapproved 2010).

Test method of viscosity, ASTM method, D4287 - 00 (Reapproved 2014).

Figure (4) I.R. Spectrum of furan-2-yl(morpholino)methanone IV

Preparation of glass panels, ASTM method, D3891 - 96

Water resistance of dried films, ASTM method, D870 – 15

Alkali resistance of dried films, Indian Standard Specification, 158(1950)

Acid resistance of dried films, Indian Standard Specification, 159(1950)

Solvent resistance of dried films, ASTM method, D2792-69 (Reapproved 2015)

Flexibility by conical mandrel bending tester, ASTM method, D522/D522M - 13

Pencil Hardness measurements, ASTM method, D3363 - 05 (Reapproved 2011) e2

Adhesion by cross-cut adhesion, ASTM method, 3359 - 09e2

28

Specular gloss, ASTM method, D523 - 14

Impact test, ASTM method, D2794 – 93 (Reapproved 2010)

Measurement of film thickness, ASTM method, D1005 – 95 (Reapproved 2013)

Preparation of steel panels for testing purpose, ASTM method, D609 – 00 (Reapproved 2012)

Corrosion Resistance test, ASTM method, D2803 – 09 (Reapproved 2015), Procedure B (ISO 4623)

Determination of degree of blistering, ASTM method, D714 – 02 (Reapproved 2009)

RESULTS AND DISCUSSION

During the course of the preparation, it was felt of interest to determine the reaction time, the reaction completion. This is indicated by watching the amount of water collected in the trap and compared its amount by that calculated. The data obtained are given in the following tables (8 and 9) and Figures (5 and 6)

 Table (8) The reaction time of formation of

 HETCA-modified alkyd resins

Dogin	Evoor	HETCA	Reaction
Cult	EXCESS-	replacement	Time
Code	OH%	(%)	(minutes)
В0		0	720
T01		5	600
T02	0%	10	300
Т03		20	90
T04		30	240
B1		0	360
T11		5	300
T12	10%	10	120
T13		20	70
T14		30	180
B2		0	480
T21		5	70
T22	20%	10	90
T23		20	50
T24		30	60
В3		0	300
T31		5	90
T32	30%	10	70
Т33		20	60
Т34		30	60



HETCA-modified alkyd resins

CA-mo	dified	alkyd	resins
Resin Code	Excess- OH%	HEFCA replacement (%)	Reaction Time (hours)
B0		0	720
F01		5	240
F02	0%	10	105
F03		20	50
F04		30	220
B1		0	360
F11		5	210
F12	10%	10	220
F13		20	210
F14		30	210
B2		0	480
F21		5	55
F22	20%	10	120
F23		20	140
F24		30	60
В3		0	300
F31		5	100
F32	30%	10	60
F33		20	55
F34		30	45

Table (9) The reaction time of formation of HEF-

The reaction rate for various resins increases by the increasing of the percentage of modifier and also increases with the increase of excess-OH percentage.

Evaluation Studies

Following the preparation of two series of



modified alkyd resins, the solid content of all formulations are adjusted to 60% solids. The

drier combinations are added as mixed combinations by 3 % from the weight of resin. The mixed driers consist of the following components:

Table (10)

Component	%
Cobalt Octoate 10%	20
Calcium Octoate 10%	45
Zirconium Octoate 12%	29
Anti skinning agent	6
Total	100

And, Varnish formula will be,

Table (11)

Component	%	
Alkyd resin or Modified alkyd resin (Solid	alkyd resin (Solid	
content=100%)	00	
Xylene	37	
Mixed Drier	3	
Total	100	

Table (12	Viceo	A	Calan	During	4	Cha	nantanistia	~ ~f	1.0	IIFT/	` A ` N/	adified	Alleria	Decimo
i abie (14	I: VISCO	ISILV.	Color.	DEVINS	ume	Ulla	racteristic	S OL	various	при	A-W	oamea	ΑΙΚΥ	i Kesins
							~								

			¥7°	ty Air			Stovi	ng dry		
		HETCA	Viscosity (mPa s)	Alr Drving	11	0 C		12	20 C	
Resin Code	Excess- OH (%)	Replacement (%)	at solid content = 70%	(HD Time) (hr)	1 hr	2 hrs	1 hr	2 hrs	Extra Stoving Drying time (hrs)	Color (Gardner)
B0		0	50	13	VST	HD	-	-	-	8
T01		5	80	10	VST	HD	-	-	-	>18
Т02	0%	10	85	> 24	ST	VST	HD	-	-	>18
Т03		20	100	> 24	Т	Т	VST	VST	3	>18
T04		30	100	> 24	Т	Т	Т	Т	>12	>18
B1		0	140	11	VST	HD	-	-	-	8
T11		5	100	8	VST	HD	-	-	-	>18
T12	10%	10	150	9	VST	HD	-	-	-	>18
T13		20	250	> 24	Т	Т	Т	Т	12	>18
T14		30	750	> 24	Т	Т	Т	Т	>12	>18
B2		0	450	7	VST	HD	-	-	-	9
T21		5	675	6	HD	-	-	-	-	>18
T22	20%	10	1100	7	VST	HD	-	-	-	>18
T23		20	4000	9	Т	ST	VST	VST	3	>18
T24		30	28500	> 24	Т	Т	ST	ST	>12	>18
B3		0	40000	5	HD	-	-	-	-	9
T31		5	144000	3.5	HD	-	-	-	-	>18
T32	30%	10	250000	4	VST	HD	-	-	-	>18
Т33		20	160000	4.5	ST	VST	HD	-	-	>18
T34		30	280000	11	Т	ST	VST	VST	3	>18

HD: Hard Dry VST: Very Slight Tacky ST: Slight Tacky T: Tacky

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Table (13): Viscosity, Color, Drying time Characteristics of various HEFCA-Modified Alkyd Resins	

							Stoving	g Dry		
				Air	110 C			120 C		
Resin Code	Excess- OH (%)	HEFCA Replacement (%)	Viscosity (mPa.s) at solid content = 70%	Drying (HD Time) (hr)	1 hr	2 hrs	1 hr	2 hrs	Extra Stoving Drying time (hrs)	Color (Gardner)
B0		0	50	13	VST	HD	-	-		8
F01		5	50	> 24	VST	HD	-	-		>18
F02	0%	10	75	> 24	VST	HD	-	-		>18
F03		20	125	> 24	Т	ST	VST	VST	>12	>18
F04		30	150	> 24	Т	Т	ST	ST	>12	>18
B1		0	140	11	VST	HD	-	-		8
F11]	5	100	3.5	HD	-	-	-		>18
F12	10%	10	150	4.0	VST	HD	-	-		>18
F13]	20	250	4.5	Т	ST	VST	HD		>18
F14		30	325	> 24	Т	Т	ST	ST	>12	>18
B2		0	450	7	VST	HD	-	-		9
F21		5	1250	4	HD	-	-	-		>18
F22	20%	10	1550	4.5	HD	-	-	-		>18
F23		20	1850	5	Т	Т	ST	ST	>12	>18
F24		30	2250	> 24	Т	Т	ST	ST	>12	>18
B3		0	40000	5	HD	-	-	-		9
F31		5	4400	4	HD	-	-	-		>18
F32	30%	10	260000	4.5	HD	-	-	-		>18
F33		20	280000	5	ST	VST	VST	VST	3	>18
F34		30	25000(at solid content =50%)	5.5	ST	VST	VST	VST	3	>18

HD: Hard Dry VST: Very Slight Tacky ST: Slight Tacky T: Tacky



Figure: 7

The drying time increase by increasing the modifier percentage and by decreasing OH-excess of alkyd resins. The drying time of 20% and 30% modification samples at low OH-excess is very bad (e.g. T03, T04, F03, F04, T14, F14).

Figure: 8

The best drying time is obtained at 30% OHexcess resins at 5% modification (e.g. T31 and F31).

-The viscosity for various resins at constant solid content increases slightly by the increas-

ing of the percentage of modifier, but increases dramatically with the increase of excess-OH percentage as shown in Figure (7 and 8)

-The color of prepared varnishes was very dark due to the presence of nitrogen atom in the modified alkyd chain.

B- Water, Acid, Alkali and Solvent Resistance

Preliminary evaluation studies are carried out to show the air drying and stoving drying films of all formulations towards water, acid, alkali and solvent resistances. The evaluation tests are conducted according to standard methods. The data are collected in tables (14 - 15). The main conclusion drown from the water, acid, alkali and solvent resistances data for the various HETCA and HEFCA-modified alkyd is tabulated in tables (10 - 11) indicate the following:

1. Except unmodified resins (B0, B01 and B03), all examined films show excellent water, acid and solvent resistance indicating no significant effect of the presence of the modifier within the employed experimental conditions.

2. The alkali resistance of the dried films is clearly improved upon the modification.

Resin	Excess-	W resis	ater stance	Acid res	sistance	Alkali R	esistance	Solvent F	Resistance
Code	OH %	А	S	А	S	А	S	А	S
B0		Ex	Ex	Ex	Ex	F	Р	Ex	Ex
T01		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T02	0%	Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T03		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T04		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
B1		Ex	Ex	Ex	Ex	Ex	F	Ex	Ex
T11		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T12	10%	Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T13		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T14		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
B2		Ex	Ex	Ex	Ex	Ex	G	Ex	Ex
T21		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T22	20%	Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T23		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T24		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
B3		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T31		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T32	30%	Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T33		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex
T34		Ex	Ex	G	Ex	Ex	Ex	Ex	Ex

Table (14): Water, Acid, Alkali and solvent Resistance data for HETCA-modified Alkyd Resins

Ex : Excellent (Almost no change) G : Good (Very slight change) F : Fair (Partially attacked) P : Poor (Complete film failure) S : Stoved drying film A: Air drying film

FORMULATION OF NEW MODIFIED ALKYD RESINS AND THEIR APPLICATION 33 Table (15): Water, Acid, Alkali and solvent Resistance data for HEFCA-modified Alkyd Resins

Resin Excess-		Water re	esistance	Acid re	sistance	Alkali R	esistance	Solvent Resistance		
Code	OH %	А	S	A	S	А	S	A	S	
B0		Ex	Ex	Ex	Ex	F	Р	Ex	Ex	
F01		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F02	0%	Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F03		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F04		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
B1		Ex	Ex	Ex	Ex	Ex	F	Ex	Ex	
F11		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F12	10%	Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F13		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F14		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
B2		Ex	Ex	Ex	Ex	Ex	G	Ex	Ex	
F21		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F22	20%	Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F23		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F24		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
B3		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F31		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F32	30%	Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F33		Ex	Ex	Ex	Ex	Ex	Ex	Ex	Ex	
F34		Ex	Ex	G	Ex	Ex	Ex	Ex	Ex	

Ex : Excellent (Almost no change)F : Fair (Partially attacked)A: Air drying film

Extensive Evaluation Studies:

Attention the directed towards more extensive studies to indicate how much the films can withstand the various attacks in terms of immersion periods. The extensive evaluation studies for water resistance, acid resistance and solvent resistance are given in table (16). The tests are carried out on the actual coating films (air dried and stoving dry) for 30 days immersion period and the films are examined for any defects.

Additionally, the extensive evaluation stud-

G: Good (Very slight change) **P**: Poor (Complete film failure)

S: Stoved drying film

ies for alkali resistance are shown in tables (17) which are carried out on the actual coating films (air dried and stoving dry) for 10 days immersion period and the films are examined for any defects.

In general, water, acid and solvent resistance for all resins are good after 30 days for both air and stoving films.

The alkali resistance of resins generally improved by increasing the modification and excess-OH% of the prepared alkyd resins.

Resin	Excess-	W resis	'ater stance	Acid Solvent resistance Resistance		Resin	Excess-	W resi	/ater stance	Acid resistance		Solvent Resistance			
Code	OH %	A S A S A	А	S	Code	OH %	Α	S	А	S	А	S			
B0		Ex	Ex	Ex	Ex	Ex	Ex	B0		Ex	Ex	Ex	Ex	Ex	Ex
T01]	Ex	Ex	Ex	Ex	Ex	Ex	F01]	Ex	Ex	Ex	Ex	Ex	Ex
T02	0%	Ex	Ex	Ex	Ex	Ex	Ex	F02	0%	Ex	Ex	Ex	Ex	Ex	Ex
T03]	Ex	Ex	Ex	Ex	Ex	Ex	F03]	Ex	Ex	Ex	Ex	Ex	Ex
T04		Ex	Ex	Ex	Ex	Ex	Ex	F04		Ex	Ex	Ex	Ex	Ex	Ex
B1		Ex	Ex	Ex	Ex	Ex	Ex	B1		Ex	Ex	Ex	Ex	Ex	Ex
T11		Ex	Ex	Ex	Ex	Ex	Ex	F11		Ex	Ex	Ex	Ex	Ex	Ex
T12	10%	Ex	Ex	Ex	Ex	Ex	Ex	F12	10%	Ex	Ex	Ex	Ex	Ex	Ex
T13		Ex	Ex	Ex	Ex	Ex	Ex	F13		Ex	Ex	Ex	Ex	Ex	Ex
T14		Ex	Ex	Ex	Ex	Ex	Ex	F14		Ex	Ex	Ex	Ex	Ex	Ex
B2		Ex	Ex	Ex	Ex	Ex	Ex	B2		Ex	Ex	Ex	Ex	Ex	Ex
T21]	Ex	Ex	Ex	Ex	Ex	Ex	F21		Ex	Ex	Ex	Ex	Ex	Ex
T22	20%	Ex	Ex	Ex	Ex	Ex	Ex	F22	20%	Ex	Ex	Ex	Ex	Ex	Ex
T23]	Ex	Ex	Ex	Ex	Ex	Ex	F23		Ex	Ex	Ex	Ex	Ex	Ex
T24		Ex	Ex	Ex	Ex	Ex	Ex	F24		Ex	Ex	Ex	Ex	Ex	Ex
B3		Ex	Ex	Ex	Ex	Ex	Ex	B3		Ex	Ex	Ex	Ex	Ex	Ex
T31		Ex	Ex	Ex	Ex	Ex	Ex	F31		Ex	Ex	Ex	Ex	Ex	Ex
T32	30%	Ex	Ex	Ex	Ex	Ex	Ex	F32	30%	Ex	Ex	Ex	Ex	Ex	Ex
T33		Ex	Ex	Ex	Ex	Ex	Ex	F33]	Ex	Ex	Ex	Ex	Ex	Ex
T34]	Ex	Ex	G	Ex	Ex	Ex	F34		Ex	Ex	G	Ex	Ex	Ex

M. M. ELSHAHAWI, et al. Table (16):Water, Acid, Alkali and solvent Resistance data for modified Alkyd Resins after 30 days

Ex : Excellent (Almost no change)G : Good (Very slight change)F : Fair(Partially attacked)P : Poor (Complete film failure)A: Air drying filmS : Stoved drying film

Table (17): Alkali and solvent Resistance data for modified Alkyd Resins after 10 days

		Alkali resis	tance (after		Excess-	Alkali resistance (after		
Resin Code	Excess-OH %	10 d	ays)	Resin Code	OH %	10 d	ays)	
		А	S		011 /0	А	S	
B0		Р	Р	B0		Р	Р	
T01		Р	Ex	F01		Р	Ex	
T02	0%	Р	Ex	F02	0%	Р	Ex	
Т03		Р	Ex	F03		Р	Р	
T04		G	G	F04		Р	Р	
B1		Р	Р	B1		Р	Р	
T11		Р	Ex	F11		F	Ex	
T12	10%	F	Ex	F12	10%	G	Ex	
T13	10%	Ex	Ex	F13		Ex	Ex	
T14		Ex	Ex	F14		Ex	F	
B2		F	Р	B2		F	Р	
T21		Ex	Ex	F21		Ex	Ex	
T22	20%	Ex	Ex	F22	20%	Ex	Ex	
T23		Ex	Ex	F23		Ex	G	
T24		Ex	Ex	F24		Ex	Р	
B3		Ex	G	B3		Ex	G	
T31		Ex	Ex	F31		Ex	Ex	
T32	30%	Ex	Р	F32	30%	Ex	Ex	
Т33		Ex	Ex	F33		Ex	F	
T34		Ex	Ex	F34		Ex	Р	

Ex : Excellent (Almost no change)G : Good (Very slight change) F : Fair (Partially attacked)P : Poor (Complete film failure)A: Air drying filmS : Stoved drying film

C- Mechanical properties of dried films:

The dry film characteristics of various modified resin films are carried out according to standard methods and well-known techniques and the results are tabulated in tables (19, 20, 21 and 22). These tables show some mechanical properties such as gloss, hardness, scratch resistance and adhesion of both varnishes and primers-containing modified alkyd resins. The formulation of primer will be as shown in the following table:

FORMULATION OF NEW MODIFIED ALKYD RESINS AND THEIR APPLICATION 35 Table (18): Zinc Phosphate Primer formula

serial	Component	%
1	Resin/modified resin	30
2	Bentone38	1
3	Ethanol	0.3
4	Wetting & dispersing Agent	0.3
5	TiO ₂	10
6	Zinc phosphate	5
7	CaCO ₃	24
8	Talc	5
9	Xylene	22.21
10	Calcium Octoate (10%)	1
11	Cobalt octoate (10%)	0.43
12	Zirconium Octoate (12%)	0.63
13	Ex skin 2	0.13
Total		100

Pagin Cada	Excess-	Film Th	ickness (j	u)	Gloss at	t 20°		Flexibility at 3 mm			Adhesion		
Kesin Code	OH %	Α	S	Р	Α	S	Р	Α	S	Р	Α	S	Р
B0		42	27	75	76	100	4	Ex	Ex	Ex	5B	5B	5B
T01		25	20	37	81	93	4	Ex	Ex	Ex	5B	5B	5B
T02	0%	25	25	57	82	85	5	Ex	Ex	Ex	5B	5B	5B
T03		35	20	45	85	93	11	Ex	Ex	Ex	5B	5B	5B
T04		30	25	36	90	95	7	Ex	Ex	Ex	5B	5B	5B
B1		25	25	42	77	96	10	Ex	Ex	Ex	5B	5B	4B
T11		40	25	43	79	90	5	Ex	Ex	Ex	4B	5B	5B
T12	10%	40	25	75	81	88	7	Ex	Ex	Ex	5B	5B	5B
T13		25	25	48	82	84	21	Ex	Ex	Ex	5B	5B	4B
T14		25	30	75	90	70	42	Ex	Ex	Ex	5B	5B	5B
B2		35	30	61	83	98	4	Ex	Ex	Ex at 5 mm	5B	5B	5B
T21		50	50	60	79	95	6	Ex	Ex	Ex	2B	5B	5B
T22	20%	70	50	65	83	67	24	Ex	Ex	Ex	5B	5B	5B
T23		55	50	50	81	89	10	Ex	Ex	Ex	2B	5B	5B
T24		60	40	50	93	89	25	Ex	Ex	Ex	5B	5B	4B
B3		35	20	56	96	99	11	Ex	Ex	Ex	0B	5B	5B
T31	1	40	45	63	89	97	9	Ex	Ex	Ex	0B	5B	5B
T32	30%	45	30	70	92	89	35	Ex	Ex	Ex at 10 mm	0B	5B	4B
T33]	35	25	60	84	76	27	Ex	Ex	Ex	0B	5B	4B
T34		30	21	70	84	92	15	Ex	Ex	Ex	0B	5B	2B

Ex: Excellent

0B : Removed area is greater than 65% , 1B : 35 – 65% removed area, 2B : 15 – 35% removed area, 3B : 5 - 15% removed area, 4B : Less than 5%, 5B: 0% removed area Table (19): Primer and Varnish Characteristic of Various HETCA-modified Alkyd Resins

Posin Codo	Excess-OH		Impact (1 K	g) Excellent at (cm)	Scratch Hardness (Pencil Hardness)				
Keshi Coue	%	Α	S	Р	А	S	Р		
B0		100	100	100	2B	8H	9H		
T01		100	100	100	В	6H	9Н		
T02	0%	100	100	100	5B	3H	9H		
Т03		100	100	100	6B	4H	8H		
T04		100	100	100	6B	6H	3B		
B1		100	100	100	2B	6H	4H		
T11		100	100	100	2B	8H	9H		
T12	10%	100	100	100	2B	6H	6H		
T13		100	100	100	2B	6H	5H		
T14		100	100	100	5B	6H	4H		
B2		100	100	100	3B	7H	9H		
T21		90	100	100	2B	HB	9H		
T22	20%	60	100	100	2B	HB	3B		
T23		100	100	100	2B	HB	5H		
T24		90	100	100	2B	HB	4H		
B3		90	100	100	7H	9H	7H		
T31		55	100	100	7H	2H	6H		
T32	30%	90	100	100	5H	2H	6H		
Т33		80	100	100	Н	F	3H		
T34		75	100	100	Н	F	3H		

6B<5B<4B<3B<2B<B<HB<F<H<2H<3H<4H<5H<6H<7H<8H<9H

Softer Harder

Table (20) Impact and Pencil hardness of HETCA-modified alkyd resin

Resin Code	Excess-OH %		Film Thi	ckness (µ)		Gloss at 60 🗆			Flexibility at 3 mm			Adhesion			
L	ا ا	A	s	Р	A	s	Р	Α	s	Р	Α	s	Р		
B0	· ,	42	27	75	76	100	4	Ex	Ex	Ex	5B	5B	5B		
F01	i r	35	25	66	79	80	14	Ex	Ex	Ex	5B	5B	5B		
F02	0%	40	30	40	79	92	9	Ex	Ex	Ex	5B	5B	5B		
F03	·	35	25	39	75	91	14	Ex	Ex	Ex	5B	5B	4B		
F04	<u> </u>	30	20	50	79	80	6	Ex	Ex	Ex	5B	5B	4B		
B1	,,	25	25	77	77	96	10	Ex	Ex	Ex	5B	5B	4B		
F11	i , , , , , , , , , , , , , , , , , , ,	45	35	40	77	73	13	Ex	Ex	Ex	5B	5B	5B		
F12	10%	50	40	34	77	75	6	Ex	Ex	Ex	5B	5B	4B		
F13	i I	40	35	66	79	84	7	Ex	Ex	Ex	5B	5B	5B		
F14	<u> </u>	40	45	60	76	91	9	Ex	Ex	Ex	5B	5B	4B		
B2	,,	35	30	61	83	98	4	Ex	Ex	Ex	5B	5B	5B		
F21	i I	65	50	44	82	72	12	Ex	Ex	Ex	0B	5B	5B		
F22	20%	45	62	60	77	87	7	Ex	Ex	Ex	0B	5B	5B		
F23	۱ ^۱	30	50	90	80	71	10	Ex	Ex	Ex	2B	5B	5B		
F24	<u> </u>	70	26	70	82	79	11	Ex	Ex	Ex	0B	5B	5B		
B3	·	35	20	56	96	99	11	Ex	Ex	Ex	0B	5B	5B		
F31	i ¹	30	20	75	91	66	6	Ex	Ex	Ex	0B	5B	5B		
F32	30%	43	30	70	86	66	7	Ex	Ex	Ex	0B	5B	5B		
F33	ı '	25	27	100	84	36	6	Ex	Ex	Ex	0B	5B	4B		
F34		15	13	56	77	77	3	Ex	Ex	Ex	2B	5B	5B		

Ex: Excellent 0B : Removed area is greater than 65%, 1B : 35–65% removed area, 2B : 15–35% removed area, 3B : 5 - 15% removed area, 4B : Less than 5%, 5B: 0% removed area Table (21): Primer and Varnish Characteristic of Various HEFCA-modified Alkyd Resins

Resin Code	Excess-OH %	Impact (1 Kg) Excellent at (cm)			Scratch Hardness (Pencil Hardness)		
		А	S	Р	Α	S	Р
B0		100	100	100	2B	8H	9H
F01		100	100	100	В	6H	9H
F02	0%	100	100	100	2B	7H	4H
F03		100	100	100	4B	7H	2B
F04		100	100	100	5B	7H	F
B1	10%	100	100	100	2B	6H	4H
F11		100	100	100	F	7H	6H
F12		100	100	100	В	3Н	4H
F13		100	100	100	В	3H	9H
F14		100	100	100	4B	2H	6H
B2		100	100	100	3B	7H	9H
F21		100	100	100	HB	6H	3H
F22	20%	100	100	100	HB	6H	2H
F23		100	100	100	Н	6H	2H
F24		100	100	100	Н	5H	9H
B3		90	100	100	7H	9H	7H
F31		35	100	100	6H	6H	6H
F32	30%	40	100	100	6H	4H	6H
F33		70	100	100	6H	3Н	7H
F34		80	100	100	Н	3Н	5H
6R<5R<4R<3R<2R <hr<f<h<2h<3h<4h<5h<6h<7h<8h<9h< th=""></hr<f<h<2h<3h<4h<5h<6h<7h<8h<9h<>							

Softer →Harder

Table (22) Impact and Pencil hardness of HEFCA-modified alkyd resins

The evaluation the corrosion resistance of prepared modified alkyd resins and corrosion inhibitors are carried out by dipping in 5% NaCL solution for 240 hours. The results are shown in the following tables:

Resin Code	Blistering		Desin Code	Blistering		
	Size	Frequency	Kesin Code	Size	Frequency	
B0	10	F	B0	10	F	
T01	10	F	F01	10	F	
T02	10	F	F02	10	F	
Т03	10	F	F03	10	F	
T04	10	F	F04	10	F	
B1	10	F	B1	10	F	
T11	10	F	F11	10	F	
T12	10	F	F12	10	F	
T13	10	F	F13	10	F	
T14	10	F	F14	10	F	
B2	10	F	B2	10	F	
T21	10	F	F21	10	F	
T22	10	F	F22	10	F	
T23	10	F	F23	10	F	
T24	10	F	F24	9	F	
B3	10	F	B3	10	F	
T31	10	F	F31	10	F	
T32	10	F	F32	10	F	
Т33	10	F	F33	10	F	
T34	10	F	F34	10	F	

 Table (23)

 Evaluation the corrosion resistance of modified alkyd resins

 Table (24): Evaluation of prepared corrosion inhibitors morpholino(thiophen-2-yl)methanone II and morpholino(furan-2-yl)methanone IV with standard commercial alkyd resin

Resin Code	Blistering		Resin Code	Blistering	
Resilf Code	Size	Frequency	Resili Code	Size	Frequency
ST	10	F	ST	10	F
ST+ 0.5% II	10	F	ST+0.5% IV	10	F
ST+ 1% II	10	F	ST+1% IV	10	F
ST+ 3% II	10	F	ST+ 2% IV	10	F
ST+ 5% II	10	F	ST+ 3% IV	10	F

 Table (25): Evaluation of prepared corrosion inhibitors morpholino(thiophen-2-yl)methanone "II" and morpholino(furan-2-yl)methanone "IV" with some prepared modified alkyd resin (synergism)

Resin Code	Blistering		Resin Code	Blistering	
	Size	Frequency	Keshi Code	Size	Frequency
T32	10	F	F32	10	F
T32+0.5% II	10	F	F32+0.5% IV	10	F
T32+1% II	10	F	F32+1% IV	7	М
T32+ 3% II	10	F	F32+2% IV	10	F
T32+ 5% II	10	F	F32+3% IV	5	MD

10 : means no blistering (the best) 0: the worst blistering degree F: few M: medium MD: medium dense The following photos of tested films after 240 hours in 5% NaCL solution are shown below:



Fig. (11): ZnPO4 primer with standard commercial alkyd resins with prepared corrosion inhibitors II & IV.

0B : Removed area is greater than 65% , 1B : 35-65% removed area, 2B : 15-35% removed area, 3B : 5 - 15% removed area,

The corrosion resistance increases by the increase of excess-OH percent of both modified and unmodified alkyd resins.

There is a certain modification percent of alkyd resins at which such alkyd resin achieves maximum corrosion resistance, and this optimum modification decreases by increasing excess-OH % of alkyd resin.

For example:

F32 + 3% IV F32 + 2% IV F32 + 1% IV F32 + 0.5% IV

Fig. (12): ZnPO4 primer with T32 and F32 alkyd resins and prepared corrosion inhibitors II and IV.

At 0% excess-OH alkyd resins (their Blank sample is B0) the maximum corrosion resistance achieves at 20- 30% modification (i.e. T03 and F04)

<u>At 10% excess-OH alkyd resins (their Blank</u> <u>sample is B01)</u> the maximum corrosion resistance achieves at 5- 20% modification (i.e. T11 and F12)

<u>At 20% excess-OH alkyd resins (their Blank</u> <u>sample is B02)</u> the maximum corrosion resistance achieves at 5- 20% modification (i.e. T23 and F23)

<u>At 10% excess-OH alkyd resins (their Blank</u> <u>sample is B01)</u> the maximum corrosion resistance

FORMULATION OF NEW MODIFIED ALKYD RESINS AND THEIR APPLICATION achieves at 5-10% modification (i.e. T32 and F31)

In other words, for achieving the maximum corrosion resistance of alkyd resins, the need for modification will decrease by increasing excess-OH percent.

The addition of prepared compounds II and IV achieves corrosion resistance effect and the optimum dosage of the addition will be as follow:

1-Morpholino(thiophen-2-yl)methanone II is 1% on total formula weight

2- Morpholino(furan-2-yl)methanone IV is 0.5% on total formula weight

The addition of prepared corrosion inhibitors II and IV on some modified alkyd resins such as T32 and F32, respectively, improves their corrosion resistance and gives additional protection (SYNERGISM) for the metal substrate at a certain dosage 0.5% on total formula weight.

Conclusions

The following observations were noticed during the investigation:

1- All resins prepared are very clear and transparent and homogenous

2- The color of modified alkyd resins is very dark due to the entering of nitrogen atom in their chains.

3- The reaction rate for various resins increases by the increasing of the percentage of modifier and also increases with the increase of excess-OH percentage.

4- The viscosity for various resins at constant solid content increases slightly by the increasing of the percentage of modifier, but increases dramatically with the increase of excess-OH percentage.

5- The drying time increases by increasing the modifier percentage and by decreasing excess-OH percent of alkyd resins. There are some difficulties in the drying of lower excess-OH % alkyd resins, and also, in higher modi-

39 fied alkyd resins due to the presence of Soya bean fatty acids in their structure.

6- All dried films are excellent water, acid, solvent resistance

7- Only the higher excess-OH percent alkyd resins are good alkali resistant.

8- The HETCA- and HEFCA-modification increases the alkali resistance of alkvd resins.

9- All films passed the adhesion and flexibility test except the higher modified alkyd resins in adhesion tests.

10-The HETCA and HEFCA modification increases the corrosion resistance performance of alkyd resins at certain optimum values.

11- The prepared Morpholino(thiophen-2-yl) methanone II and Morpholino(furan-2-yl)methanone IV are corrosion inhibitors additives and they can improve the corrosion resistance of HET-CA- and HEFCA- modified alkyd resins.

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