

12-1-2014

Section: Chemistry

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Abd El-Ghaffar, Heba; Bakr, Wafaa; and Abou Shahba, Rabab (2014) "EXTRACTION OF RADIUM ISOTOPES FROM SOME COASTAL SEAWATER SAMPLES IN EGYPT METHOD VALIDATION," *Al-Azhar Bulletin of Science*: Vol. 25: Iss. 2, Article 16.

DOI: <https://doi.org/10.21608/absb.2014.23796>

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## **EXTRACTION OF RADIUM ISOTOPES FROM SOME COASTAL SEAWATER SAMPLES IN EGYPT METHOD VALIDATION**

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### **ABSTRACT**

*A radiochemical separation of Ra isotopes was carried out to determine the activity levels of radium (<sup>226</sup>Ra and <sup>228</sup>Ra) in seawater samples, as well as providing a base-line data on background radiation in the investigated locations. Seawater samples were collected from selected sites along the coasts. The water samples were analyzed for radium isotopes (Ra-226 & Ra-228) using alpha and liquid scintillation detectors. Development steps were applied to the radioanalysis procedure by using calcium phosphate co-precipitation and the addition of isopropanol. Quality assurance and method validation of the developed procedure was carried out to assure the results. The obtained results were within the uncertainty level  $\pm 5\%$ . Quality control steps were applied through the efficiency calibration of the detectors, the estimation of uncertainties, the use of blanks and the comparison with other laboratories. For comparison, selected samples were measured by gamma spectrometry using high purity germanium detector, after radiochemical separation of the isotopes with ion-exchange chromatography using a strong cation exchange resin, and the activity concentrations of <sup>228</sup>Ra were comparable with those analyzed by the LSC. The average activity concentrations of <sup>226</sup>Ra and <sup>228</sup>Ra in the Mediterranean seawater samples were found to be 166.8 and 500.6 mBq/L respectively, while in the Red seawater samples were 13.0 and 176.7 mBq/L respectively. The maximum value of radium activity was found at RasElBarr estuary, where the River Nile fresh water mixes with the coastal seawater. On the basis of the current results, we may conclude that the radium activities in the investigated seawater samples are well comparable to the corresponding reported values in literature.*

*Keywords: radium, seawater, alpha spectrometry, method validation, Mediterranean.*

### **1. INTRODUCTION**

Determination of radium in environmental water has become a matter of interest in public health because radium is one of the most hazardous elements with respect to internal exposure. Radium is of particular interest for oceanographers as it may lead to an understanding of various mobilization processes in the ocean, and studies of isotopic concentrations and ratios in the ocean can provide information about mixing processes and groundwater discharge into the ocean [1]. According to several investigators, the <sup>226</sup>Ra/<sup>228</sup>Ra activity ratio in waters is extremely variable [2]. Several studies have shown that Ra isotope relationships are suitable for determination of fluxes and mixing rates of continental waters into the oceans and estuaries, and exchange between ground water and surface water [3].

The main feature of this northern coast is the presence of a large number of beaches and resorts. The Red Sea tourism is largely dependent on the surrounding environment such as sand, water quality, and coral reefs. On the Red Sea coast, there are oil exploration and production

plants as well as phosphate ore mining and production, which are non-nuclear pollution sources that could have serious radiological impacts on the coastal environment of the Red Sea. Radionuclides are known to be associated with organic materials in nature. Therefore, oil, gas and oil field brines frequently contain radioactive materials, which may increase the risk of radiation exposure to the environment [4]. Phosphate deposits are generally characterized by enhanced radionuclide concentrations, consequently, the mining and processing of this phosphate ore redistribute radionuclides throughout the environment. Considering popularity of these beaches and the habits of the population and tourists to spend long hours on the beach, it was a useful approach to estimate the radioactivity in the seawater of these areas. Therefore, the aim of this study is to determine the activity concentrations of radium isotopes (<sup>226</sup>Ra and <sup>228</sup>Ra) in selected locations along the Egyptian coastal areas of the Mediterranean and Red sea.

### **2. METHODOLOGY**

Unless otherwise indicated, all references to water refer to deionized water (DDW).

### 2.1 Sampling

The collection of samples was performed at selected sampling points along the coasts. The selection of sampling sites was based primarily on the population existence at these beaches.

The seawater samples were collected from six locations along the Mediterranean coast from Rasheed to Port Said. For the red sea coast, nine samples were collected along the coast from AinSokhna to Hurghada, as shown in fig (1,2). The seawater samples were collected from the surface, at a distance of 30 meters from the shore. Five liters from each sample were collected, filtered out through a membrane of diameter =  $0.45\mu$ , acidified with 11 M HCl solution at the rate of 10 ml per liter of sample immediately. After filtration to avoid the adsorption of radionuclides on the walls of the container, transferred to polyethylene bottles and transferred to laboratory.

### 2.2 Materials and Apparatus

Radium extraction from water samples were carried out using Diphonix Resin (50-100 and 100-200 mesh), supplied from Triskem International, 35170 Bruz, France. Standard reference solutions of  $^{226}\text{Ra}$  and  $^{228}\text{Ra}$  were supplied by the

National Institute of Standards and Technology (NIST), (SRM 4967A, SRM 4339B). The  $^{133}\text{Ba}$  standard solution was supplied by North American Technical Services (NATS) (EZ-83879-767). The cation exchange resin was used in a column mode with BioRad Glass Econo columns of 0.9 cm diameter, together with polypropylene funnels and Teflon end fittings connected with plastic taps. All gamma radioactivity measurements were carried out using a Canberra HPGe coaxial detector with relative photo-peak efficiency of 40% for the 1332 keV line of  $^{60}\text{Co}$ . The germanium detector was connected to a 8192 multichannel analyzer with counting capacity of 228 counts per channel. The alpha spectrometric analysis were carried out using a Canberra Alpha Analyst, with a chamber containing a passivated implanted planar silicon (PIPS) detector with an active area of  $450\text{ mm}^2$ . The efficiency of the detector was calibrated against a standard alpha multi-source (67970-121, Analytics Co.) using the certified activity of the measured radionuclides. The beta measurements of  $^{228}\text{Ac}$  were carried out using Liquid Scintillation detector, Packard, model Tri-Carb 2770 TR/SL. All other chemicals used in this study, including  $\text{KMnO}_4$ , isopropanol, ammonium sulphate and different mineral acids were of analytical grade.

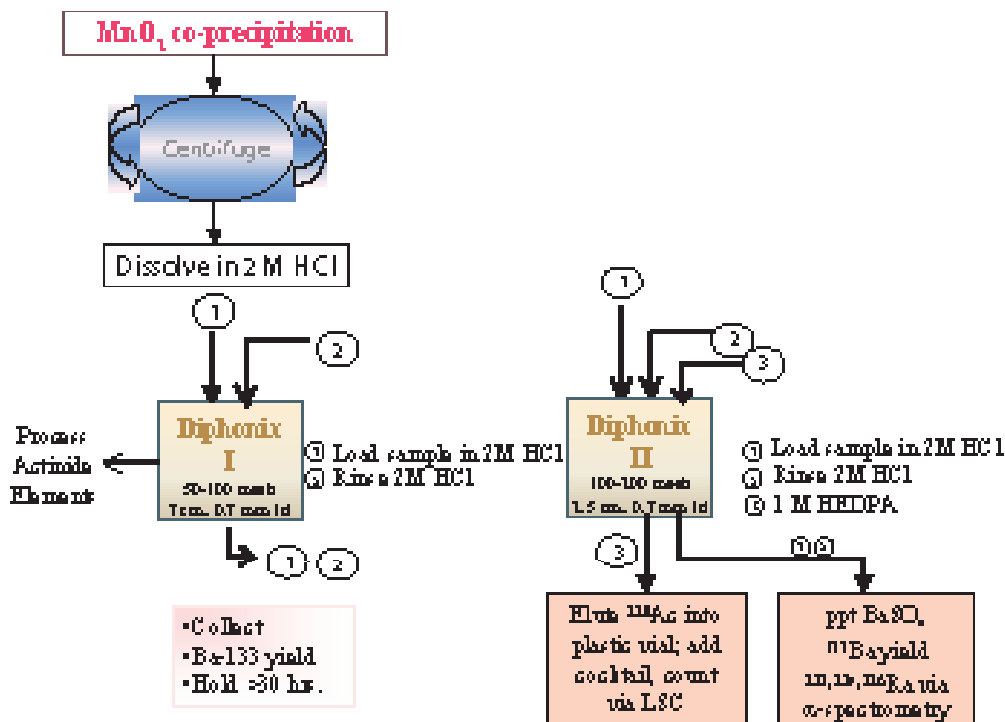


Fig.(2-1): The location of seawater samples along the Mediterranean and red sea coasts

### 2.3 Radioanalyses and Measurements

The radium isotopes ( $^{226}\text{Ra}$  and  $^{228}\text{Ra}$ ) were determined in the water samples following a combined alpha and beta spectrometric method described by S. Nour et al, 2004 [5]. In this approach,  $^{226}\text{Ra}$ ,  $^{228}\text{Ra}$  and  $^{133}\text{Ba}$  tracer are co-precipitated with  $\text{MnO}_2$ , dissolved in 2M HCl, loaded into a Diphonix resin column to eliminate other interfering radionuclides and is then kept for  $^{228}\text{Ac}$  ingrowth period of two days, reloaded again into

a second diphonix resin column. The collected Ra/Ba fraction was precipitated using  $\text{BaSO}_4$  micro-precipitation. During the barium sulphate micro-precipitation, the formation of white  $\text{BaSO}_4$  precipitate was observed. Development steps were carried out by using calcium phosphate co-precipitation and adding  $(\text{NH}_4)_2\text{SO}_4$  and isopropanol to the barium sulphate micro-precipitation in a tube, mixed well and allowed to stand in an ice bath for 30 min. before vacuum filtration[6], as shown in the flowchart.



The filter was mounted on a disk, counted by gamma for the  $^{133}\text{Ba}$  recovery and the  $^{226}\text{Ra}$  is assessed by alpha spectrometry. The  $^{228}\text{Ac}$ , held in the column, was eluted with HEDPA (1M 1-Hydroxyethane-1,1-diphosphonic acid) in a LS vial, cocktail was added and the vial was counted. The chemical recovery was determined from the  $^{133}\text{Ba}$  yield using the 276, 302 and 384 keV energy lines by gamma spectrometry.

### 2.4 Quality Control and Method Validation

Validation steps

One of, or a combination of, the following:

- calibration using reference standards or reference materials;
- comparison of results achieved with

other methods;

- interlaboratory comparisons;
- systematic assessment of the factors influencing the result;
- assessment of the uncertainty of the results based on scientific understanding of the theoretical principles of the method and practical experience.

For quality assurance and validation purpose, blank samples were prepared in the same manner as the corresponding samples, and measured for background estimation, and detection of any radioactive contamination. Reference water samples were determined using the same analysis and measure-

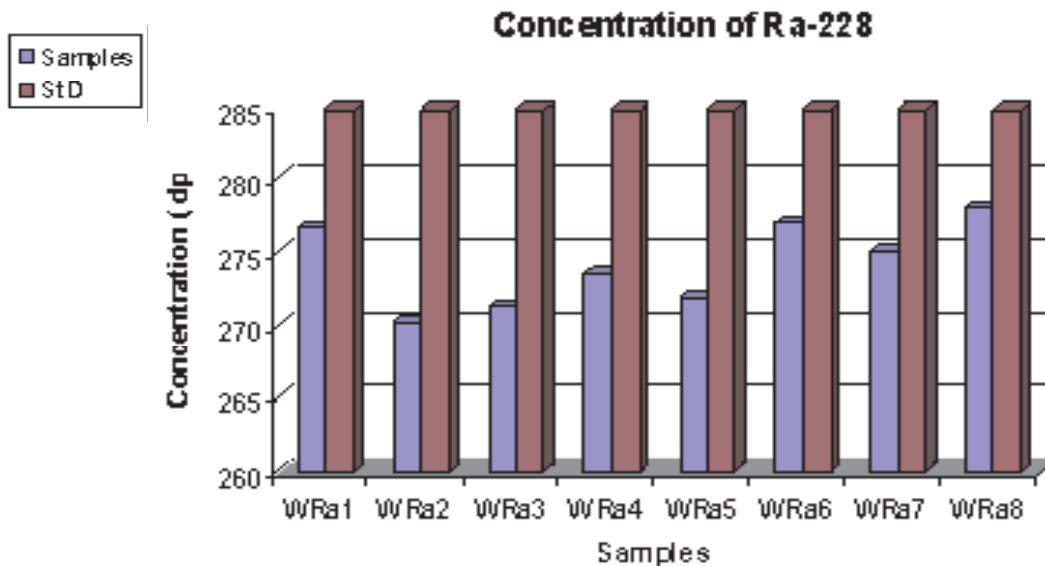
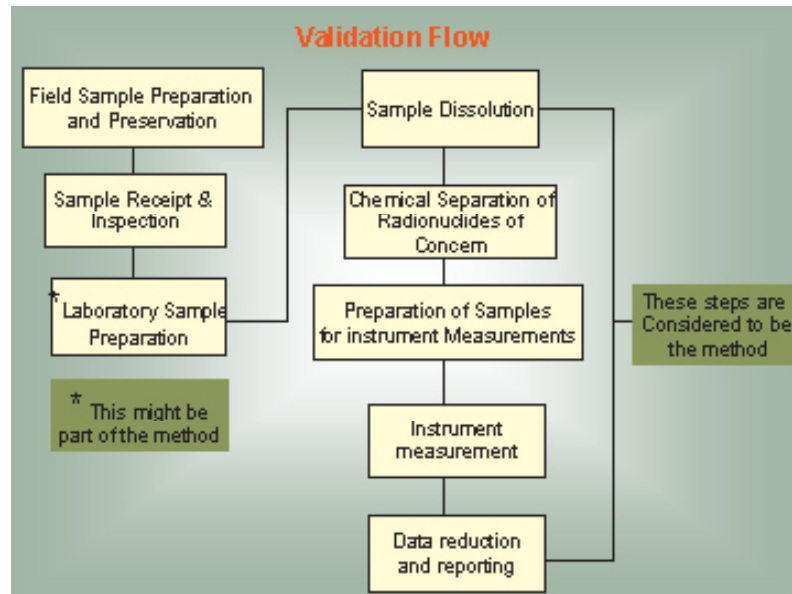


Fig. (2-2) : Activity concentrations of Ra-228 compared with the standard in the water samples

ment protocol, and were compared against their certified values to test the closeness of the measured samples to its reference values. To verify the validity of the final results,

spiked water samples were analyzed for <sup>228</sup>Ra by following the method described in flowchart to monitor any defects in the cali-

Table (2-1) : Results of Ba-133 yield, Ac-228 net cpm and Ra-228 activity

Samples	Ba yield as a fraction	Net cpm of Ac-228	Concentration of Ra-228 (dpm)	Chemical Recovery
WRa1	0.970	211.0	276.7 ± 1.6	97.0 %
WRa2	0.975	214.6	270.2 ± 1.2	94.8 %
WRa3	0.984	193.4	271.3 ± 0.9	96.2 %
WRa4	0.975	204.3	273.6 ± 1.0	96.0 %
WRa5	0.980	213.3	271.9 ± 1.4	95.4 %
WRa6	0.970	208.9	277.0 ± 1.6	97.2 %
WRa7	0.974	204.2	275.1 ± 1.3	96.5 %
WRa8	0.982	215.3	278.0 ± 1.5	97.5 %



brations, prepared solutions and measurements. Table (2-1) and fig.(2-2) represent the resulted <sup>228</sup>Ra activities.

Some selected samples were analyzed for radium isotopes (<sup>226</sup>Ra and <sup>228</sup>Ra), the procedure described by A.El-Sharkawy et al, 2013 [7], where the seawater samples were allowed to pass through strong cation exchange resin (purolite) in a column mode with BioRad Glass Econo columns, the resin was transferred to standard counting containers and the containers were tightly sealed for four weeks to allow secular equilibrium between <sup>226</sup>Ra, <sup>228</sup>Ra and their decay products. The efficiency calibration of the germanium detector for the radium isotopes (Ra-226 and Ra-228) was carried out using standard samples. Known activity resins were prepared by spiking water (DDW) samples with known amounts of <sup>226</sup>Ra and <sup>228</sup>Ra. The spiked resin samples containing a known amount of the radionuclides of interest were used to provide a identical matrix with a known activity, and all other conditions were followed typically (flow rate, resin volume,

counting time and geometry). The <sup>226</sup>Ra activities were determined via its daughters <sup>214</sup>Pb and <sup>214</sup>Bi through the gamma energy lines 295.22, 351.93 and 609.31 keV. The <sup>228</sup>Ra activities were determined through the gamma energy lines of 338.32 and 911.2 keV. The calculated specific activities were basically performed using a comparison method:

$$A_{unk} = \frac{A_{std}}{CR_{std}} \cdot CR_{unk}$$

where;

A<sub>unk</sub> is the calculated activity of the sample;

A<sub>std</sub> is the activity of the standard resin;

CR<sub>std</sub> is the counting rate for the standard resin; and

CR<sub>unk</sub> is the counting rate of the unknown sample.

**Validation through proficiency test**

The validation was carried out through the participation in two proficiency testing exer-

**Table (2-2) : Results of the IAEA water analysis**

Sample code	Recovery	Combined standard uncertainty %	IAEA reference value (Bq / l)	Our values (Bq / l)
IAEA-423	88.0 %	0.78	8.4	6.8
IAEA-426	74.0 %	0.56	6.5	5.9
IAEA-427	90.0 %	0.70	65.0	63.1
IAEA-428	81.0 %	0.35	3.6	3.3
IAEA-431	90.0 %	2.4	23.9	22.8

**Table 3-1: The radium activities in the Mediterranean and Red sea water samples**

Locations	TDS	Ra-226		Ra-228	
I- The Mediterranean	g/L	mBq/L	±	mBq/L	±
Rasheed (W1)	32.4	293.5	12.7	766.4	28.2
Borollos (W2)	36.4	58.2	4.1	386.8	14.4
Balteem (W3)	33.9	129.0	9.0	564.2	28.2
Gamasa (W4)	37.1	108.5	7.3	195.1	8
Dametta (W5)	35.1	69.5	4.9	238.7	10.0
RasAlBarr (W6)	30.3	328.6	16.4	1156.4	64.8
Port Said (W7)	40.7	180.3	12.6	196.4	8.8
II- The Red sea					
Ain Sokhna1 (A)	38.8	6.8	0.5	107.7	5.4
Ain Sokhna2 (H)	45.5	23.9	1.7	134.5	6.7
Zafarana (J)	43.1	12.1	1.0	102.0	5.1
Zafarana 2 (K)	45.5	11.9	0.8	211.2	9.7
RasGareb (L)	42.8	11.6	0.9	70.5	4.6
Rasshokair (O)	44.7	6.3	0.6	269.3	14.8
Gulf of Jimsah1 (P)	44	6.0	0.5	317.6	15.9
Gulf of Jimsah2 (R)	45.2	26.1	1.8	297.5	15
Hurghada (T)	41.5	12.7	0.8	80.3	4.9

cises. The results of the water analysis are summarized in Table (2-2). From the results, it is concluded that the applied method of analysis is “fit for purpose” to measure radium in various waters (low and very high salinity).

Errors were propagated due to nuclear counting statistics ( $\sigma_N$ ), tracer ( $\sigma_s$ ) and volume ( $\sigma_v$ ), and the combined total uncertainty was calculated according to the following equation:

$$\sigma_T = \sqrt{(\sigma_N)^2 + (\sigma_V)^2 + (\sigma_S)^2}$$

**MDA:** The minimum detectable activity at 95% confidence level is given by;

$$MDA = \frac{2.71 + 4.65\sqrt{B}}{T \times \text{eff} \times Y \times V}$$

Where, T : counting time

B : background counts

eff : efficiency

Y : Ba-133 yield

V : volume of samples in liters.

The minimum detectable activity (MDA) was calculated according to the equation presented by Currie, 1968 [8], and it was 6.6 mBq and 1.2 mBq for the beta and alpha measurements respectively.

### 3. RESULTS AND DISCUSSION

In the following section, the results of the radium levels in the Mediterranean and Red sea water samples will be represented.

#### *Radium Isotopes in Water Samples*

The activity concentrations of  $^{226}\text{Ra}$  and  $^{228}\text{Ra}$  in the Mediterranean and Red seawater samples

are represented in table 3-1.

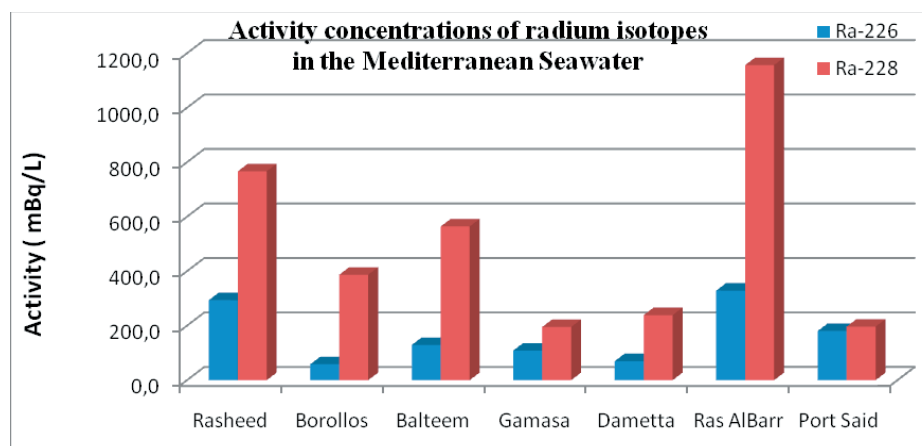
The average activity concentration of  $^{226}\text{Ra}$  in the Mediterranean water sample was 166.8mBq/L, (ranging from 58.2to 328.6mBq/L), while the average activity concentration of  $^{228}\text{Ra}$  was 500.6mBq/L, (ranging from 195.1 to 1156.4mBq/L), as shown in figure 3-1. The activity concentrations of radium in the Mediterranean seawater samples obtained are comparable to those reported in literature. The Suez Canal water samples showed an average activity concentrations of 0.6 and 0.5 Bq/L for  $^{226}\text{Ra}$  ( $^{238}\text{U}$ ) series and  $^{228}\text{Ra}$  ( $^{232}\text{Th}$  series) respectively [9].

In a radiological survey along the Mediterranean coast from Alexandria to Rasheed, the activity concentrations of Ra-226 (U-series) and Ra-228(Th-232 series) in Rasheed coastal seawater were 1.9 and 0.5 Bq/L respectively [10].

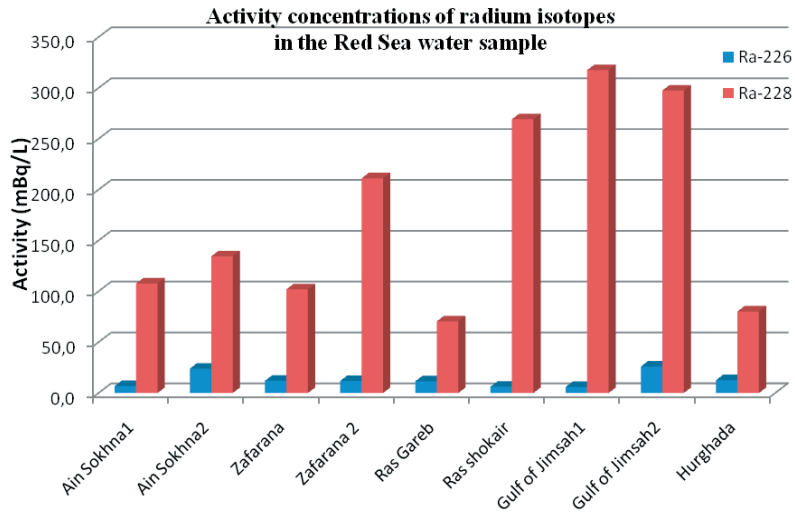
The sample W6, which was collected from RasAlBarr estuary, showed relatively the maximum radium activity. This may be attributed to the desorption of radium from suspended particles and bottom sediments, as the River Nile fresh water meets and mixes with the coastal seawater [11, 12].

The average activity concentration of  $^{226}\text{Ra}$  in the Red Sea water sample was 13.0mBq/L, (ranging from 6.0to 26.1 mBq/L), while the average activity concentration of  $^{228}\text{Ra}$  was 176.7mBq/L, (ranging from 70.5 to 317.6mBq/L), as shown in figure 3-2.

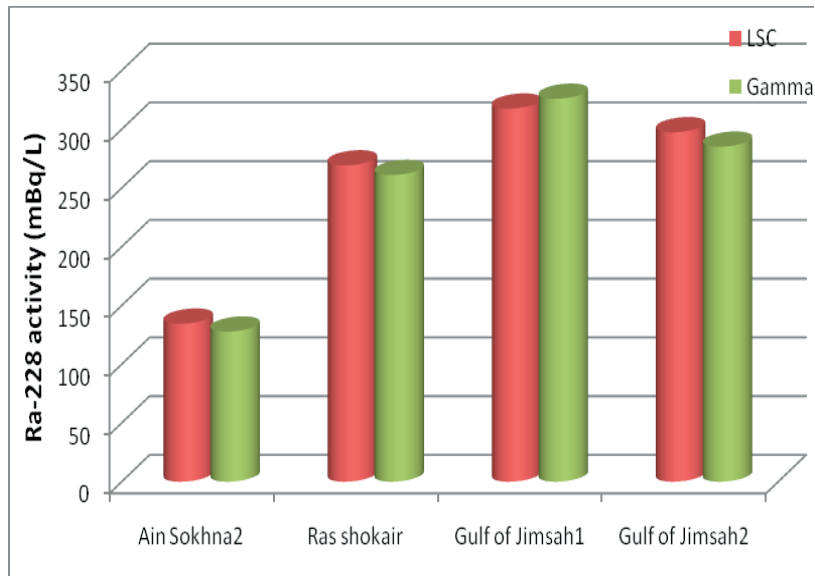
The sea water samples collected from RasShokair and GabalZait (O,P, R locations)



**Fig 3-1 :Radium Concentrations in the Mediterranean Seawater**



**Fig 3-2 :Radium Concentrations in the Red Seawater**



**Fig 3-3 :comparison of <sup>228</sup>Ra activities by gamma and LSC**

showed relatively the maximum radium activities, which may be explained by the presence of Ras el Behar field, one of the most important oil exploration and production-area along the Red Sea coast, at about few kilometers from their locations [13].

Selected samples were analyzed by gamma spectrometry for comparison with other laboratory, and the <sup>226</sup>Ra activity concentrations were within the low limit of detection of the germanium detector at the 609.3keV energy line (1.5 mBq/L). The <sup>228</sup>Ra activity

concentrations were comparable with those analyzed by LSC, as shown in fig.(3-3).

#### 4. CONCLUSION

Surveillance monitoring along the Mediterranean and Red sea coasts has been carried out to determine the radium activity levels in sea water, and providing a base-line data on radium activities in the coastal seawaters. Radium isotopes (<sup>226</sup>Ra and <sup>228</sup>Ra) were analyzed in theseawater samples from different locations, and the average activity concentrations were found to be 166.8 and 500.6mBq/L for <sup>226</sup>Ra and <sup>228</sup>Rarespectively



in the Mediterranean seawater samples, while the average activity concentrations of  $^{226}\text{Ra}$  and  $^{228}\text{Ra}$  in the Red seawater samples were 13.0 and 176.7 mBq/L respectively. The maximum value of radium activity was found at RasEl-Barr estuary, where the River Nile fresh water mixes with the coastal seawater. Some of the Red sea water samples were analyzed by different techniques using gamma spectrometry for comparison, and the activity concentrations of  $^{228}\text{Ra}$  were comparable with those analyzed by the LSC. On the basis of the current results, we may conclude that the radium activities in the investigated seawater samples are well comparable to the corresponding reported values in literature.

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