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PHYSICO-CHEMICAL AND MECHANICAL PROPERTIES OF BLENDED CEMENT PASTES CONTAINING RICE HUSK ASH AND METAKAOLIN

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ABSTRACT

*This work aims to study the effect of partial substitution of ordinary Portland cement (OPC) by rice husk ash (RHA) and metakaolin (MK) on the physico-chemical and mechanical properties of the hardened OPC-RHA-MK blended cement pastes. OPC was partially replaced by different ratios of MK (10, 15 and 20%) and a constant ratio of RHA (5%) and the resulted cement blends were hydrated in the paste form by using the water/cement ratios required for the standard water of consistency; the pastes thus obtained, were hydrated for 1, 3, 7, 28, and 90 days. At the end of hydration period, the cement pastes were tested for compressive strength, total porosity and hydration kinetic via determination of free lime contents. The phase composition of the formed hydration products was investigated using X-ray diffraction (XRD) and differential thermal analysis (DTA) techniques. It was found that, the substitution of ordinary Portland cement (OPC) by rice husk ash (5% RHA) enhances the physico-chemical and mechanical properties of the hardened blended cement pastes as compared with the neat OPC. The results of compressive strength indicated slightly higher values for the pastes made of OPC-RHA-MK blends containing 5% RHA blended with 10, 15 % MK. However, the blended cement paste derived from OPC-RHA-MK blend containing 5% RHA blended with 20% MK, for economic reasons, was taken as the most suitable mix containing both RHA and MK. The partial substitution of OPC by RHA and MK leads to higher porosity values with a consequent decrease in the compressive strength values especially during the early ages of hydration. It was found that, the increase of MK content in OPC-RHA-MK blended cement pastes resulted in an increase in water consistency and setting times. Lower values of free lime contents were obtained for OPC-RHA-MK blended cement pastes, with the formation of further additional amounts of CSH, as a result of the pozzolanic reaction. **Keywords:** Blended Cement, Rice husk ash, Metakaolin, DTA, XRD.*

1. INTRODUCTION

Portland cement concrete is the most widely used human-made commodity on the planet; about 25 billion metric tons are produced globally each year^[1]. Global Portland cement production currently accounts for 7% (2×10^9 tones) of anthropogenic carbon dioxide (CO_2) emissions annually, resulting mainly from production of cement clinker, the active binding ingredient of concrete^[2]. Because kiln-fired Portland cement is an energy-intensive material, requiring 4–5 GJ per ton of cement^[3]. About half of these emissions occur through combustion of fossil fuels. The remaining emissions result from calcination of limestone: one kg of Portland cement clinker releases 0.87 kg of CO_2 to the atmosphere^[4]. At present, efforts have been made to promote the use of pozzolans to partially replace Portland cement. Pozzolana is a natural or artificial material containing silica in a reactive form. A more formal definition of ASTM C618^[5], describes pozzolana as a siliceous or siliceous and aluminous

material which in itself possesses little or no cementitious value but will, in finely divided form and in the presence of moisture, chemically react with calcium hydroxide at ordinary temperatures to form hydrated cementitious properties^[6,7]. In recent years, many researchers have established that the use of pozzolanic materials, like blast furnace slag, silica fume, metakaolin (MK), fly ash (FA) and rice husk ash (RHA) etc., can not only improve the various properties of concrete, but also contribute to economy in construction costs^[8]. Rice husks, sometimes called rice hulls, are one of the major agricultural by-products and are the shells produced during the dehusking operation of paddy rice. It constitutes 20% of the 500 million tons of paddy produced in the world^[9]. The use of RHA decreases the demand for cement in the construction industry, reduces the cost of concrete production, and reduces the negative environmental impact that CO_2 emissions represent in the production of cement. The ash produced by controlled burning of the

rice husk between 550°C and 700°C incinerating temperature for 1h transforms the silica content of the ash into non-crystalline or amorphous silica ^[10]. Metakaolin (MK) is one of the pozzolanic materials that have been most studied in recent times. MK is an artificial pozzolana obtained from the calcination of kaolinitic clays at temperatures around 700–850 °C. Due to its high pozzolanic activity, the inclusion of MK improves the mechanical properties and durability of concrete ^[11].

2. EXPERIMENTAL

The materials used in this study were ordinary Portland cement (OPC), rice husk ash (RHA) and Metakaolin (MK). OPC was supplied from Arabian Cement Company, Al Mosalah Portland Cement CEM I- 42.5N; Egypt, Rice husk ash (RHA) was obtained by burning of rice husk in a muffle furnace under controlled burning condition (a heating rate 10°C/min at 650°C for 3 h), followed by cooling to room temperature in desiccator and ground to pass 90 µm sieve. Metakaolin (MK) was prepared by calcination of kaolinite clay which was supplied from Middle East Mining Investments Company, Egypt. Kaolinite clay was calcined in a muffle furnace with a heating rate 10 °C/min at 750°C for 3 h, to give metakaolin (MK) ; metakaolin recharged from the muffle furnace, cooled to room temperature in a desiccator and ground to pass 90 µm sieves. The results of chemical analysis of these materials are shown in Table (1).

Table (1): Chemical oxide composition of starting materials, (wt., %).

| Oxide contents,(%) | OPC | RHA | MK |
|--------------------------------|-------|-------|-------|
| SiO ₂ | 19.94 | 74 | 61.04 |
| Al ₂ O ₃ | 4.46 | 0.79 | 35.57 |
| Fe ₂ O ₃ | 3.6 | 0.27 | 0.34 |
| CaO | 65.91 | 0.84 | 0.44 |
| SO ₃ | 2.73 | 0.68 | 0.24 |
| MgO | 1.67 | 1.57 | ---- |
| Na ₂ O | 0.33 | ---- | ---- |
| K ₂ O | 0.05 | 5.27 | 0.04 |
| TiO ₂ | ---- | ---- | 1.18 |
| P ₂ O ₅ | ---- | 12.10 | 0.33 |
| L.O.I | 1.31 | 5.04 | 0.51 |

Different dry mixes were prepared by the partial substitution of OPC by different ratios of MK (10, 15 and 20%) and constant ratio of RHA (5%). Each dry mix was blended in a porcelain ball mill using three balls for six hours in order to attain a complete homogeneity, then kept in airtight containers. The mix composition of the different OPC-RHA-MK cement blends are given in Table (2).

The water of consistency and setting times (initial and final) of the fresh cement pastes, made of each cement blend, were determined using Vicatapparatus according to ASTM: C191^[12]. To prepare the blended cement pastes made of OPC-RHA-MK blends, an amount of each cement blend was placed on a smooth, non-absorb-

Table (2): Mix composition of the various blended cements as well as the optimum water of consistency and setting times of their fresh pastes.

| Mixes | OPC, (%) | RHA, (%) | MK, (%) | W/C Ratio | Setting times, (min) | |
|-------|----------|----------|---------|-----------|----------------------|-------|
| | | | | | Initial | Final |
| B | 100 | 0 | 0 | 0.3 | 77 | 170 |
| OR2 | 95 | 5 | 0 | 0.33 | 114 | 234 |
| ORM1 | 85 | 5 | 10 | 0.335 | 138 | 265 |
| ORM2 | 80 | 5 | 15 | 0.345 | 140 | 275 |
| ORM3 | 75 | 5 | 20 | 0.355 | 147 | 290 |

bent surface, and a crater was formed in the center. The required amount of mixing water (water of consistency) was poured into the crater by the aid of a trowel. The dry cement was slightly trowel over the remaining to absorb the water for about one minute. Continuous and vigorous mixing was made for three minutes then the mixing was completed. The fresh blended cement paste was placed into $2.5 \times 2.5 \times 2.5$ cm cubic moulds, manually pressed into the corners and along the surface of the mould until a homogeneous paste was obtained. After the top layer was compacted and pressed with hand, the surface of the paste was smoothed by the aid of thin edged trowel. Immediately after moulding, the specimens were first cured in a humidity chamber at 100 % R.H. at room temperature $23 \pm 1^\circ\text{C}$ for 24 hours. At the end of the moist curing period, the cubic specimens were demoulded and curing was continued under tap water up to 3, 7, 28 and 90 days, ASTM: C191^[12]. At each hydration time, all pastes were tested for their compressive strength, total porosity, free lime and phase composition of the formed hydration products. A set of three cubic specimens representing the same cement paste and curing time were used for the determination of compressive strength of the hardened paste according to ASTM C-150^[13] and the average value was recorded. The total porosity of the hardened cement pastes was determined according to Copeland and Hayes^[14].

After the compressive strength determination, removal of free water was accomplished by using a stopping solution from 1:1 mixture by volume of methyl alcohol and acetone. At any time a representative sample of cement paste, about 10 g, was taken and ground in alumina mortar under the surface of the stopping solution (100 ml), then filtered through sintered glass funnel (G4). Washing the content of the funnel was carried out using 50 ml of fresh diethyl ether and dried at 70°C in the drying furnace then kept in airtight bottles.

The free lime content, CaO (%), was determined by using the glycerol/ethanol extraction method and the mean value of the two independent determinations were recorded, Kondo et al.^[15].

Differential thermal analysis (DTA) was carried out using the dried specimens; the apparatus used in the present study was DTA-50 thermal analyzer (Schimadzu Co. Tokyo, Japan). A sample of about 50 mg was used with heating rate for adjusted at $20^\circ\text{C}/\text{min}$. under dynamic nitrogen atmosphere. X-ray diffraction technique was carried out on some selected hydrated cement pastes using a Philips diffractometer with a scanning speed of $20^\circ\text{C}/\text{min}$. and Ni-filtered Cu-K_α radiation. The identification of all samples was confirmed by computerized research of the PDF data obtained from the Joint Committee on Powder Diffraction Standards-International Center for Diffraction Data (JCPDS-ICDD), 2001.

3. RESULTS AND DISCUSSION

3.1. Water of consistency and setting time

The water of consistency and setting times (Initial and final) of the OPC-RHA-MK blended cement pastes are shown in Table (2). It was found that the paste made of OPC-RHA-MK mix which is presented as ORM3 (containing 5 % RHA and 20 % MK) possesses the highest water of consistency and setting times (Initial and final) values. The increase in water demand could be attributed to the high fineness of metakaolin and RHA as well as their narrow particle size distribution. The initial and final setting times of OPC-RHA-MK blends were affected by the metakaolin content. For cements with 20% metakaolin content there was a delay in the setting process^[16].

3.2. Compressive strength

The results of compressive strength of the hardened OPC-RHA-MK blended cement pastes, made with and without the partial replacement of OPC by 5% RHA (OR2) and different percentages of replacement of OPC by MK (10, 15 and 20 %) are graphically plotted as a function of curing time up to 90 days in Fig. (1). The results of Fig. (1) show that the compressive strength of the hardened pastes made of all mixes increases continuously with increasing age of hydration. This increase in strength is mainly attributed to the formation and later accumulation of hydration products within the pore system of hardened pastes. At the early ages of hydration (up to 7 days), it is clear that the best mixes are ORM1

and ORM2 made with OPC replacement of 5 % RHA together with 10 or 15 % MK, respectively; the pastes made of these two mixes possess almost the same compressive strength values as compared to the neat OPC and higher compressive strength as compared to those of OPC-RHA paste (mix OR2). Evidently, the strength results at the early age of hydration can be attributed to the two effects which connected to the development of strength of OPC-RHA-MK blended cement at early ages; these are, (i) the acceleration of OPC hydration, which occurs within the first 24h, and (ii) the pozzolanic reaction, which has its maximum effect during the later ages of hydration for all MK levels between 5% and 20%^[17]. Metakaolin had a very positive effect on the cement strength especially at the later hydration ages. This may be due to the fact that produced metakaolin as well as the commercial one gave similar hydration products after 28 days and the pozzolanic reaction was accelerated between 7 and 28 days^[16]. The hydration products of both OPC and OPC-RHA-MK blends, mainly as calcium silicate hydrates, represent the main binding centers between the remaining unhydrated parts of OPC and artificial pozzolana (MK and RHA) grains. The main conclusions derived from the results of compressive strength obtained are summarized as follows:

(i) The blended cement derived from OPC-RHA mix OR2 represents the most suitable mix containing RHA as a partial replacement of OPC.

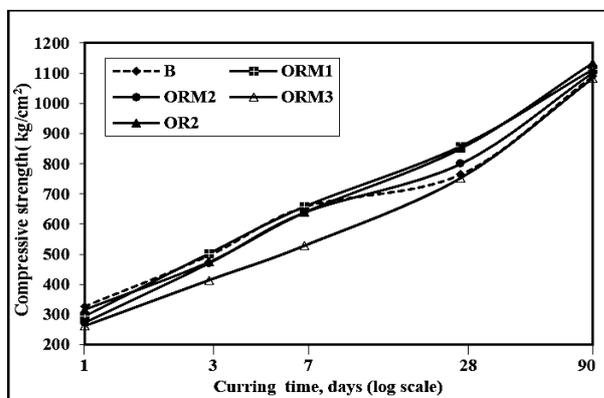


Fig.(1):Compressive strength (kg/cm²) of blended cement pastes made of the neat OPC,OPC-RHA and OPC-RHA-MK blends as a function of curing time.

(ii) The blended cement derived from OPC-RHA-MK mix ORM3, for economic reasons, the most suitable mix containing both RHA and MK as partial replacements of OPC.

These two conclusions are based on the basis of the reasonable strength values of the hardened blended cement pastes made of OR2 and ORM3 mixes; these values are slightly lower than those of the OPC at the early ages of hydration, while at the later ages of hydration, higher values of strength are obtained for these blended cement pastes as compared to those of the neat OPC paste.

3.3. Total porosity

The incorporation of RHA and/or MK as partial replacements of OPC resulted in a very dense microstructure of the hardened blended cement pastes with lower total porosity in comparison with the neat OPC paste^[18]. The results of total porosity of hardened blended cement pastes, made of OPC-RHA of mix OR2 with different replacements by MK (10, 15 and 20%), at the various ages of hydration up to 90 days are graphically plotted as a function of curing time in Fig. (2).It is clear that the total porosity of the hardened cement pastes made of all mixes decreases with the increase of curing time due to the progress of hydration which leads to filling of large fraction of the total pore system of the hardened cement pastes. Obviously, the results of Fig (2) show that the total porosity increases with the increase the level of OPC replacement by MK up to 7 days of hydration compared to OPC paste with and without RHA; because the total pore volume decreases with an increasing rate up to 14 days^[19]. The incorporation of MK to OPC-RHA blend (Mix OR2) enhances the pozzolanic reaction with the free CH liberated from OPC hydration especially at the early ages of hydration. At the longer ages of hydration (28-90 days), the hardened pastes made of all mixes of OPC-RHA with and without MK exhibit lower porosity values than those of the neat OPC paste.

3.4. Free lime content:

Kinetics of hydration was studied by the determination of free lime (CaO, %) at the different

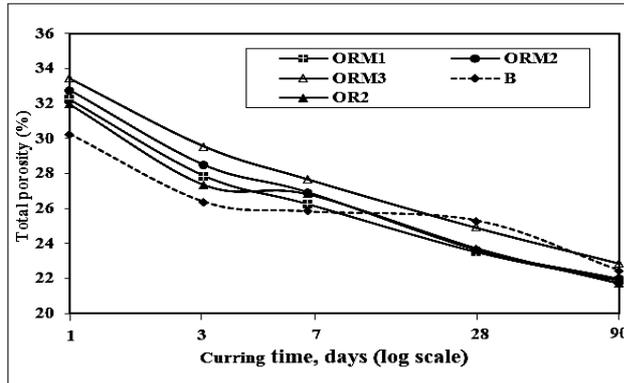


Fig. (2): The total porosity of the hardened blended cement pastes made of the neat OPC, OPC-RHA and OPC-RHA-MK blends at the different curing ages (days).

ages of hydration. Evidently, the hydration of the neat OPC paste is associated with a continuous liberation of free portlandite (CH) from the hydration of OPC clinker phases [20]. The free lime contents obtained for the hardened blended cement pastes made from OPC-RHA blend (OR2) with different percentages of replacement of OPC by MK (10, 15 and 20%) were determined at the different ages of hydration up to 90 days and these are graphically plotted as a function of curing time in Fig. (3). It is clear from the results of Fig. (3) that, the free lime contents (CaO, %) obtained for the various blended cement pastes, made of OPC-RHA and OPC-RHA-MK mixes, increase with increasing age of hydration up to 3-days; this is followed by a gradual decrease with increasing age up to 90-days of hydration. The values of the free lime content obtained for the various OPC-RHA-MK blended cement pastes represent a net effect of the amount of free lime liberated during OPC hydration and the free lime consumed as a result of the pozzolanic reaction between CH and RHA and/or MK. At the early ages of hydration (up to 3 days), the free lime liberated from OPC fraction exceeds the free lime consumed by the pozzolanic reaction; while at the later ages of hydration, however, the amount free lime consumed by the pozzolanic reaction exceeds the amount of free lime liberated during OPC hydration.

3.5. Differential thermal analysis (DTA)

The DTA thermograms obtained for the hardened of OPC (B), OPC-RHA-MK mixes (ORM1,

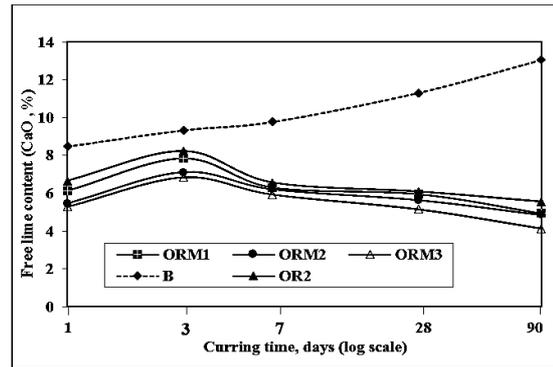


Fig. (3): Free lime content of the hardened blended cement pastes made from the neat OPC, OPC-RHA and OPC-RHA-MK blends as a function of curing ages time.

ORM2 and ORM3) and OPC-RHA mix (OR2) after 7 and 90 days of hydration are shown in Figs. (4-7). The first endothermic peak located at about 80-120°C characterizes the removal of free water and the decomposition of the nearly amorphous calcium silicate hydrates, mainly as CSH(I) and CSH(II), as well as calcium sulphoaluminate hydrates. The second endothermic peak observed at about 140-170°C represents the decomposition of calcium aluminate hydrates, mainly as C_4AH_{13} , as well as the crystalline CSH. The third endothermic peak located at about 440-470°C is characteristic for the decomposition of CH. The last two endothermic peaks appeared at 700-740 and 750-810 °C represent the decomposition of amorphous $CaCO_3$ and its crystalline form, respectively. The endotherms characteristic for calcium silicate hydrates (CSH) becomes more distinguishable for the hardened OPC-RHA-MK blended cement pastes (ORM1, ORM2 and ORM3) specially after 7-day of hydration and extended to the later age of hydration (90-days) which exhibit a marked decrease in the intensity of portlandite peak and increasing intensities of CSH peaks; this is due to the pozzolanic reaction of the amorphous silica in RHA and amorphous silica and alumina in MK with the free lime released from OPC hydration. These results are in agreement with the data obtained for the free lime content reported before in this investigation. This larger consumption of CH is also confirmed from the results of free lime content which is strongly reduced at the lat-

er ages of hydration. In conclusion, the addition of blending materials such as pozzolana (RHA and MK) can effectively reduce the CH content of the hardened OPC paste.

3.6. X-ray diffraction (XRD) analysis:

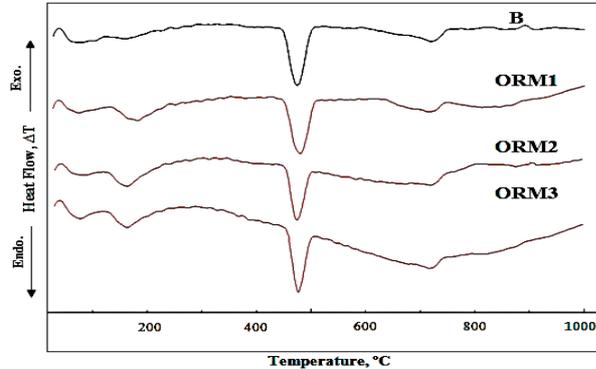


Fig. (4): DTA thermograms obtained for the hardened cement pastes made of the neat OPC and OPC-RHA-MK blends at 7 days of hydration.

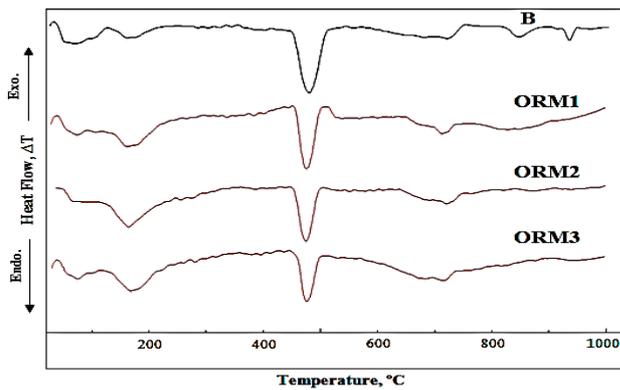


Fig. (5):DTA thermograms obtained for the hardened cement pastes made of the neat OPC and OPC-RHA-MK blends at 90 days of hydration.

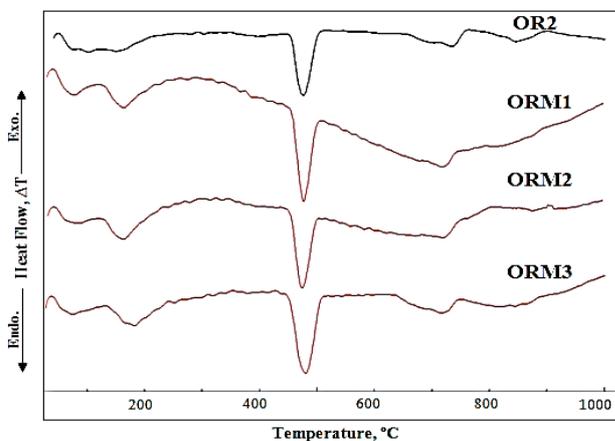


Fig. (6): DTA thermograms obtained for the hardened cement pastes made of mixes OPC-RHA and OPC- RHA-MK blends at 7 days of hydration.

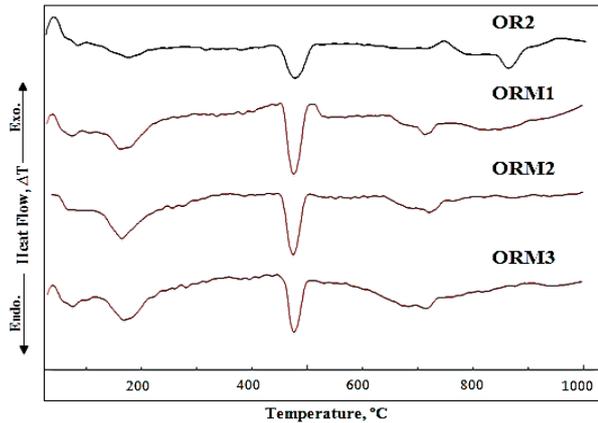


Fig. (7): DTA thermograms obtained for the hardened cement pastes made of mixes OPC-RHA and OPC-RHA-MK blends at 90 days of hydration.

The XRD patterns obtained for the hardened pastes, made of the neat OPC (mix, B), OPC substituted with 5% rice husk ash (mix, OR2) and OPC substituted with 5%RHA and different ratios of MK (mixes, ORM1, ORM2 and ORM3), after 7 and 90 days of hydration are shown in Figs. (8-11). Obviously, the XRD patterns shown in Figs. (8-11) indicate the presence of the main hydration products which are calcium silicate hydrates (CSH), calcium aluminosilicate hydrates (CASH) and calcium hydroxide (CH), in addition to the remaining unhydrated parts of OPC clinker phases (C_3S and $\beta-C_2S$) as well as the quartz phase in case of the hardened OPC-RHA-MK blended cement pastes. In addition, CASH and CAH phases were observed for the OPC-RHA-MK pastes where MK appeared to have a catalyzing effect on OPC hydration, leading to an acceleration in the reaction rates; an increase in cumulative heat evolved was observed in a previous publication during early hydration, and for some cements apparently an increased intensity in heat evolved during certain periods of early hydration [21]. However, the results show a gradual decrease of peak intensities characteristic for CH, accompanied by a corresponding increase of the peak intensities characteristic for calcium silicate hydrates (CSH), was observed in the XRD diffractograms obtained for OPC-RHA-MK blended cement pastes; this increase is related to the amount of MK in OPC-RHA-MK blends pastes during the first 7-days of hy-

dration. The results indicate also that MK has a fast pozzolanic activity at early ages of hydration; it does consume higher amounts of CH liberated from the hydration of OPC at early ages of hydration. At the later ages of hydration, the peaks of the anhydrous cement components (C_3S and $\beta-C_2S$) are almost disappeared. Since there is a clear decrease of the (CH) content at the longer ages of hydration (beyond 3 days), it is concluded that it was consumed in its pozzolanic interaction with RHA and MK. This is also confirmed by the decrease in intensities of the main CH peaks, especially after 90-days of hydration. This is attributed to the formation of additional amounts of calcium silicate hydrates (CSH-II) and calcium aluminate hydrates (CAH).

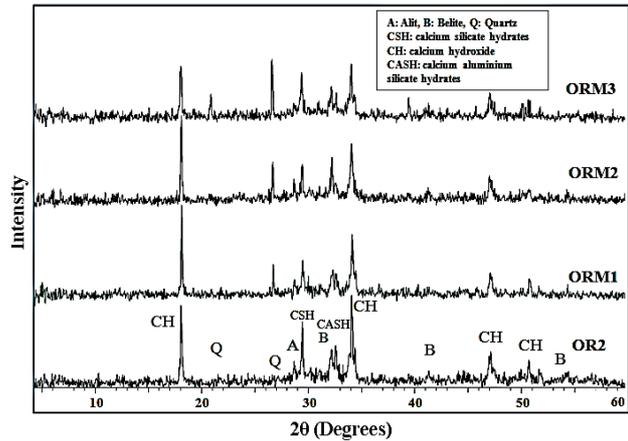


Fig. (10): The XRD patterns obtained for the hardened cement pastes made from the mixes OR2, ORM1, ORM2 and ORM3 at 7-days of hydration.

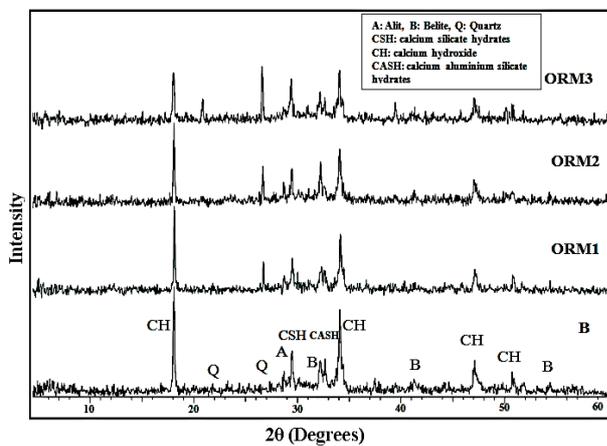


Fig. (8): The XRD patterns obtained for the hardened cement pastes made from the neat OPC and mixes OR2, ORM1, ORM2 & ORM3 at 7-days of hydration.

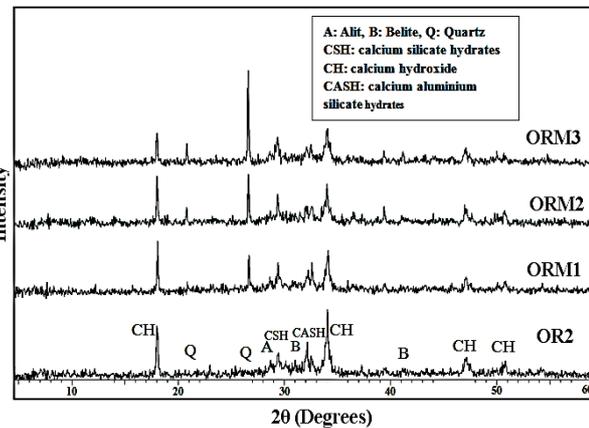


Fig. (11): The XRD patterns obtained for the hardened cement pastes made from the mixes OR2, ORM1, ORM2 and ORM3 at 90-days of hydration.

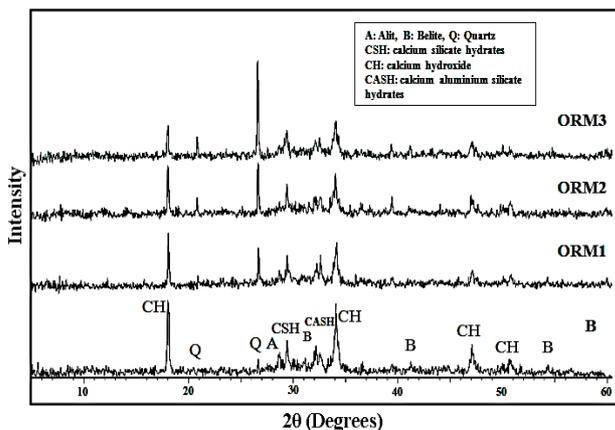


Fig. (9): The XRD patterns obtained for the hardened cement pastes made from the neat OPC and mixes OR2, ORM1, ORM2 & ORM3 at 90-days of hydration.

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